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TITLE:

Preparation of thiobenzimidazole derivatives as

chymase inhibitors

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PATENT ASSIGNEE(S): Teijin Limited, Japan SOURCE: PCT Int. Appl., 103 pp.

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$$R^{1}$$
 N
 $T-CH_{2}-A-E$
 $G-J$

AB The title compds. I [T = S(0)m; R1, R2 = H, halo, etc.; A = single bond, etc.; E = CO2R3, etc.; R3 = H, alkyl; G = alkylene; further details on G are given; m = 0 - 2; J is, for example, aryl, etc.; extensive details on J are given] are prepd. Compds. of this invention in vitro showed IC50 values of 10 nM to 100 nM against chymase. A formulation is given.

IT 255397-52-5P 255397-58-1P 255397-59-2P

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255397-98-9P

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. of thiobenzimidazole derivs. as chymase inhibitors)

RN 255397-52-5 CAPLUS

CN Acetic acid, [[5,6-dimethyl-1-(phenylmethyl)-1H-benzimidazol-2-yl]thio](9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{Me} \\ \\ \text{Me} \\ \\ \text{CH}_2\text{-Ph} \end{array}$$

RN 255397-58-1 CAPLUS

CN Butanoic acid, 4-[[5,6-dimethyl-1-[2-(1-naphthalenyl)ethyl]-1H-benzimidazol-2-yl]thio]- (9CI) (CA INDEX NAME)

Me N S-
$$(CH_2)_3$$
- CO_2H
N CH2
CH2

RN 255397-59-2 CAPLUS

CN Butanoic acid, 4-[[1-[2-(1-naphthalenyl)ethyl]-1H-benzimidazol-2-yl]thio](9CI) (CA INDEX NAME)

RN 255397-98-9 CAPLUS

CN Butanoic acid, 4-[[5,6-dimethyl-1-(1-naphthalenylmethyl)-1H-benzimidazol-2-yl]sulfonyl]- (9CI) (CA INDEX NAME)

Me N S (CH₂)
$$_3$$
 - CO₂H O CH₂

REFERENCE COUNT:

9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT



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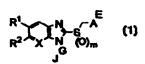
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(54) THIOBENZIMIDAZOLE DERIVATIVES

(57) Thiobenzimidazole derivatives represented by general formula (1) or pharmaceutically acceptable salts thereof which have a potent human chymase inhibitory activity and, therefore, are usable as clinically applicable preventives and/or remedies for various diseases in which human chymase participates.



***** (40)。

Description

Technical Field

The present invention relates to thiobenzimidazole derivatives represented by the formula (1) and, more specifically, thiobenzimidazole derivatives useful as inhibitors of human chymase activity.

Background Art

[0002] Chymase is one of the neutral proteases present in mast cell granules, and is deeply involved in a variety of biological processes in which mast cells participate. Various effects have been reported including, for example, the promotion of degranulation from mast cells, the activation of interleukin-1β (IL-1β), the activation of matrix protease, the decomposition of fibronectin and type IV collagen, the promotion of the release of transforming growth factor-β (TGF-β), the activation of substance P and vasoactive intestinal polypeptide (VIP), the conversion of angiotensin I (Ang I) to Ang II, the conversion of endothelin, and the like.

[0003] The above indicates that inhibitors of said chymase activity may be promising as preventive and/or therapeutic agents for diseases of respiratory organs such as bronchial asthma; inflammatory/allergic diseases, for example allergic rhinitis, atopic dermatitis, and urticaria; diseases of circulatory organs, for example scierosing vascular lesions, intravascular stenosis, disturbances of peripheral circulation, renal failure, and cardiac failure; diseases of bone/cartilage metabolism such as rheumatoid arthritis and osteoarthritis, and the like.

[0004] As inhibitors of chymase activity, there are known triazine derivatives (Japanese Unexamined Patent Publication (Kokai) No. 8-208654); hydantoin derivatives (Japanese Unexamined Patent Publication (Kokai) No. 9-31061); imidazolidine derivatives (PCT Application WO 96/04248); quinazoline derivatives (PCT Application WO 97/11941); heterocyclic amide derivatives (PCT Application WO 96/33974); and the like. However, the structures of these compounds are entirely different from those of the compounds of the present invention.

[0005] On the other hand, an art related to the compounds of the present invention is disclosed in U.S. Pat. No. 5,124,336. Said specification describes thiobenzimidazole derivatives as having an activity of antagonizing thromboxane receptor. The specification, however, makes no mention of the activity of said compounds to inhibit human chymase.

20 [0006] Thus, it is an object of the present invention to provide novel compounds that are potential and clinically applicable inhibitors of human chymase.

Disclosure of the Invention

Thus, after intensive research to attain the above objective, the applicants of the present invention have found the following 1 to 15 and have thereby completed the present invention.

1. A thiobenzimidazole derivative represented by the following formula (1):

wherein,

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R¹ and R², simultaneously or independently of each other, represent a hydrogen atom, a halogen atom, a trihalomethyl group, a cyano group, a hydroxy group, an alkyl group having 1 to 4 carbons or an alkoxy group having 1 to 4 carbons, or R¹ and R² together form -O-CH₂-O-, -O-CH₂-CH₂-O- or -CH₂-CH₂-CH₂-, in which the carbons may be substituted with one or a plurality of alkyl groups having 1 to 4 carbons; A represents a single bond, a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, a substituted or unsubstituted arylene group having 6 to 11 carbons, or a substituted or unsubstituted

heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, in which the substituent represents a halogen atom, OH, NO₂, CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a linear or branched alkylthio group having 1 to 6 carbons, a linear or branched alkylsulfonyl group having 1 to 6 carbons, a linear or branched acyl group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a trihalomethyl group, a rihalomethoxy group, a phenyl group, an oxo group, or a phenoxy group that may be substituted with one or more halogen atoms, and in which the substituents may be independently substituted at any one or more sites of the ring or the alkylene group;

E represents COOR³, SO₃R³, CONHR³, SO₂NHR³, a tetrazole group, a 5-oxo-1,2,4-oxadiazole group or a 5-oxo-1,2,4-thiadiazole group in which R³ represents a hydrogen atom, or a linear or branched alkyl group having 1 to 6 carbons:

G represents a substituted or unsubstituted, linear or branched alkylene group having 1-6 carbons that may be interrupted with one or a plurality of O, S, SO_2 , and NR^3 , in which R^3 is as defined above and the substituent represents a halogen atom, OH, NO_2 , CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a trihalomethyl group, a trihalomethoxy group, a phenyl group, or an oxo group; m represents an integer of 0 to 2;

when m is 0 and A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, then J represents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted aryl group having 7 to 9 carbons, a substituted aryl group having 10 to 11 carbons, a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring;

when m is 0 and A is a substituted or unsubstituted arylene group having 6 to 11 carbons or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, then J represents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted aryl group having 6 to 11 carbons, or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring; or

when m is 0 and A is a single bond or when m is 1 or 2, then J represents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted aryl group having 6 to 11 carbons, or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, in which the substituent represents a halogen atom, OH, NO₂, CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a linear or branched alkylthio group having 1 to 6 carbons, a linear or branched acyl group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a substituted or unsubstituted anilide group, a trihalomethyl group, a trihalomethoxy group, a phenyl group, an oxo group, a COOR³ group, or a phenoxy group that may be substituted with one or more halogen atoms, and in which the substituents may be independently substituted at any one or more sites of the ring or the alkylene group; and

X represents CH or a nitrogen atom;

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or a medically acceptable salt thereof (hereinafter referred to as "the thiobenzimidazole derivative of the present invention").

- 2. The thiobenzimidazole derivative characterized in that, in the above formula (1), A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, a substituted or unsubstituted arylene group having 6 to 11 carbons, or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
- 3. The thiobenzimidazole derivative characterized in that, in the above formula (1), A is a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
- 4. The thiobenzimidazole derivative characterized in that, in the above formula (1), m is 1, or a medically acceptable salt thereof.
- 5. The thiobenzimidazole derivative characterized in that, in the above formula (1), m is 2, or a medically acceptable salt thereof.
- 6. The thiobenzimidazole derivative characterized in that, in the above formula (1), m is 0, A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, and J is a substituted or unsubstituted aryl

group having 7 to 9 carbons, a substituted anyl group having 10 to 11 carbons, or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.

- 7. The thiobenzimidazole derivative characterized in that, in the above formula (1), m is 0, A is a substituted or unsubstituted arylene group having 6 to 11 carbons or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, and J is a substituted or unsubstituted aryl group having 6 to 11 carbons or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
- 8. The thiobenzimidazole derivative characterized in that, in the above formula (1), G is -CH₂-, -CH₂-CH₂-, -CH₂CO-, -CH₂CO-, -CH₂CO-, -CH₂CONH-, -CO-, -SO₂-, -CH₂SO₂-, -CH₂S-or -CH₂CH₂S-, or a medically acceptable salt thereof.
- 9. The thiobenzimidazole derivative characterized in that, in the above formula (1), R¹ and R² simultaneously represent a hydrogen atom, a halogen atom, an alkyl group having 1 to 4 carbons or an alkoxy group having 1 to 4 carbons, or R¹ and R², independently of each other, represent a hydrogen atom, a halogen atom, an alkyl group having 1 to 4 carbons, an alkoxy group having 1 to 4 carbons, a trihalomethyl group, a cyano group, or a hydroxy group, or a medically acceptable salt thereof.
- 10. The thiobenzimidazole derivative characterized in that, in the above formula (1), E represents COOH or a tetrazole group, or a medically acceptable salt thereof.
- 11. The thiobenzimidazole derivative characterized in that, in the above formula (1), X represents CH, or a medically acceptable salt thereof.
- 12. A thiobenzimidazole derivative characterized by having an activity of inhibiting human chymase, or a medically acceptable salt thereof.
- 13. A pharmaceutical composition comprising an at least one thiobenzimidazole derivative or a medically acceptable salt thereof and a pharmaceutically acceptable carrier.
- 14. A pharmaceutical composition which is a preventive and/or therapeutic agent for a disease.
- 15. A preventive and/or therapeutic agent wherein said disease is an inflammatory disease, an allergic disease, a disease of respiratory organs, a disease of circulatory organs, or a disease of bone/cartilage metabolism.
- 30 Best Mode for Carrying Out the Invention

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[0008] The present invention will now be explained in more detail below.

[0009] The above definitions concerning the substituents of the compounds of formula (1) of the present invention are as follows:

[0010] R¹ and R², simultaneously or independently of each other, represent a hydrogen atom, a halogen atom, a trihalomethyl group, a cyano group, a hydroxy group, an alkyl group having 1 to 4 carbons or an alkoxy group having 1 to 4 carbons, or R¹ and R² together form -O-CH₂-O-, -O-CH₂-CH₂-O- or -CH₂-CH₂-CH₂-CH₂-, in which the carbons may be substituted with one or a plurality of alkyl groups having 1 to 4 carbons. As the alkyl group having 1 to 4 carbons, there can be mentioned a methyl group, an ethyl group, a (n, i-) propyl group and a (n, i, s, t-) butyl group, and preferably a methyl group may be mentioned. Preferably R¹ and R² simultaneously represent a hydrogen atom, a halogen atom, an alkyl group having 1 to 4 carbons or an alkoxy group having 1 to 4 carbons, or an alkoxy group having 1 to 4 carbons. As the halogen atom, a hydroxy group, an alkyl group having 1 to 4 carbons, or an alkoxy group having 1 to 4 carbons. As the halogen atom, as used herein, there can be mentioned a fluorine atom, a chlorine atom, a bromine atom and the like, and preferably a chlorine atom and a fluorine atom may be mentioned. As the alkyl group having 1 to 4 carbons, there can be mentioned a methyl group, an ethyl group, an ethyl group, and preferably a methyl group may be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group having 1 to 4 carbons, there can be mentioned. As the alkoxy group and a (n, i, s, t-) butyloxy group, and preferably a methoxy group may be mentioned.

[0011] A represents a single bond, a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, a substituted or unsubstituted arylene group having 6 to 11 carbons, or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring. Preferably, there can be mentioned a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, a substituted or unsubstituted arylene group having 6 to 11 carbons, or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring. As the substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, there can be mentioned a methylene group, an ethylene group, a (n, i-) propylene group and a (n, i, t-) butylene group, and preferably an ethylene group may be mentioned. As the substituted or unsubstituted arylene group having 6 to 11 carbons, there can be mentioned a phenylene group, an indenylene group and a naphthylene group etc., and preferably a phenylene group

may be mentioned. As the substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, there can be mentioned a pyridilene group, a furanylene group, a thiophenylene group, an imidazolene group, a thiazolene group, a pyrimidilene group, an oxazolene group, an isoxazolene group, a benzphenylene group, a benzimidazolene group, a quinolilene group, an indolene group, a benzothiazolene group and the like, and preferably a pyridilene group, a furanylene group, and a thiophenylene group may be mentioned.

[0012] Furthermore, as the substituent, as used herein, there can be mentioned a halogen atom, OH, NO₂, CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons in which the substituent may be joined to each other at adjacent sites via an acetal bond, a linear or branched alkylthio group having 1 to 6 carbons, a linear or branched acyl group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a trihalormethyl group, a trihalormethoxy group, a phenyl group, or a phenoxy group that may be substituted with one or more halogen atoms. They may be independently substituted at any one or more sites of the ring or the alkylene group, a phenyl group, a bromo group, a nitro group, a methoxy group, a cyano group, a methylenedioxy group, a trifluoromethyl group, a methyl group, an ethyl group, a (n, i, t-) butyl group, and the like.

[0013] As E, there can be mentioned COOR³, SO₃R³, CONHR³, SO₂NHR³, a tetrazole group, a 5-oxo-1,2,4-oxa-diazole group or a 5-oxo-1,2,4-thiadiazole group, and preferably COOR³ or a tetrazole group may be mentioned. As R³ as used herein, there can be mentioned a hydrogen atom or a linear or branched alkyl group having 1 to 6 carbons, and preferably a hydrogen atom, a methyl group, an ethyl group, or a t-butyl group may be mentioned, and most preferably a hydrogen atom may be mentioned.

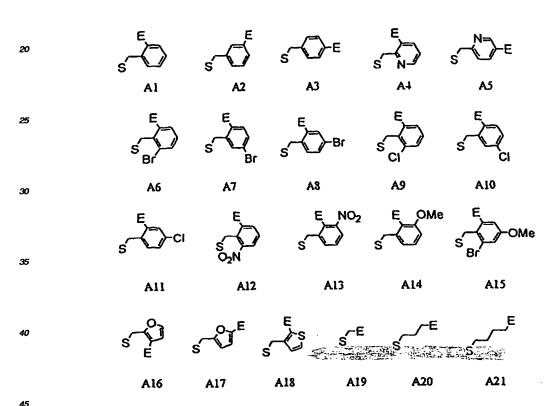
[0014] ©represents a substituted or unsubstituted, linear or branched alkyl group having 1 to 6 carbons that may be interrupted with one or a plurality of O, S, SO₂, and NR³, in which R³ is as defined above and the substituent represents a halogen atom, OH, NO₂, CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a tri-halomethyl group, a trihalomethoxy group, a phenyl group, or an oxo group. specifically, there can be mentioned -CH₂-, -CH₂CH₂-, -CH₂CO-, -CH₂CO-, -CH₂CONH-, -CO-, -SO₂-, -CH₂SO₂-, -CH₂S-, -CH₂CH₂S- and the like, and preferably -CH₂-, -CH₂CH₂-, -CH₂CO- or -CH₂CH₂O- may be mentioned.

[0015] m represents an integer of 0 to 2, and preferably 0 or 2 may be mentioned.

When m is 0 and A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, then trepresents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted anyligroup thaving 7 to 9 carbons, a substituted anyligroup having 10 to 11 carbons, a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring. Preferably, a substituted aryl group having 10 to 11 carbons and a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring may be mentioned. As the substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, there can be mentioned a methyl group, an ethyl group, a (n, i-) propyl group, a (n, i, s, t-) butyl group, a (n, i, ne, t-) pentyl group and a cyclohexyl group. As the substituted or unsubstituted anyl, group having 7 to 9 carbons, there can be mentioned an indenyl group, and as the substituted anyl group having 10 to 11 carbons, there can be mentioned anaphthyligroup. As the substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, there can be mentioned a pyridyl group, a furanyl group, a thiophenyl group, an imidazole group, a thiazole group, a pyrimiding group; an oxazole group; an isoxazole group, a 🛒 👊 benzofurane group, a benzimidazole group, a quinoline group, an isoquinoline group, a quinoxaline group, a benzoxadiazole group, a benzothiadiazole group, an indole group, a N-methylindole group, a benzothiazole group, a benzothiophernyl group, a benzisoxazole group and the like, and preferably a benzothiophenyl group or a N-methylindole group may be mentioned.

a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a linear or branched alkylthio group having 1 to 6 carbons, a linear or branched alkylsulfonyl group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a substituted or unsubstituted anilide group, a trihalomethyl group, a trihalomethoxy group, a phenyl group, or a phenoxy group that may be substituted with one or more halogen atoms. They may be independently substituted at any one or more sites of the ring or the alkyl group. Specifically, there can be mentioned OH, a chloro group, a bromo group, a nitro group, a methoxy group, a cyano group, a methylenedioxy group, a trifluoromethyl group, a trifluoromethyl group, a methyl group, an ethyl group, a (n, i, s, t-) butyl group, an anilide group and the like.

- 10 [0018] X represents CH or a nitrogen atom, and preferably CH may be mentioned.
 [0019] As the compound of formula (1), specifically those described in Tables 1 to 40 are preferred, Most preferred among them are compounds Nos. 37, 50, 63, 64, 65, 84, 115, 117, 119, 121, 123, 130, 143, 147, 168, 174, 256, 264, 272, 311, 319, 320, 321, 324, 349, 352, 354, 355, 358, 364, 380, 392, 395, 398, 401, 402, 444, 455, 459, 460, 506, 863, 866, and 869.
- 15 [0020] A1 to A21 and J1 to J85 described in Tables 1 to 40 are the groups shown below, in which E and G are as described above.



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5	O _c	\bigcirc G	\bigcirc _G	(C)c	$_{F}\bigcirc^{G}$	Ç, G	\bigcirc_{G}^{F}
	n	12	13	J4	J 5	36	J7
10	cl Q e	CI	$\mathcal{O}_{\mathbf{c}}^{cl}$	Br G	Ģ ^G Br	\mathcal{O}_{g}^{g}	
	J8	J9	J10	nı	Jl2.	J(3	
15	F ₃ C G	CF ₃	CC _{CF3}	MeO G	OMe	OMe	
20	314	J15	J16	J17	J18	J19	
	F ₃ CO G	OCF ₃	OCF3	NC OG	ÇN CN	(CCN CN	
25	J20	J21	722	J23	J24	J25	
30	EIO CG	OEt	OEt	O ₂ N O ^G	NO ₂	O _{NO2}	
	J26	J27	J28	J29	130	131	
35	но	OH G	\mathcal{O}_{e}^{oH}	Q ^c	T,	A.	
40	J32	J33	J34	135	J36	J37	
~	Çc	CI CI CI	cı Ca	CI CI	CI CI CI	C G	
45	138	139	140	J41	142	J43	
50	MeO OMe	G OMe OMe	MeO G OMe	MeO CG	MeO OM	G G	Me OMe
	344	J45	J46	J47	J48	J49	J50

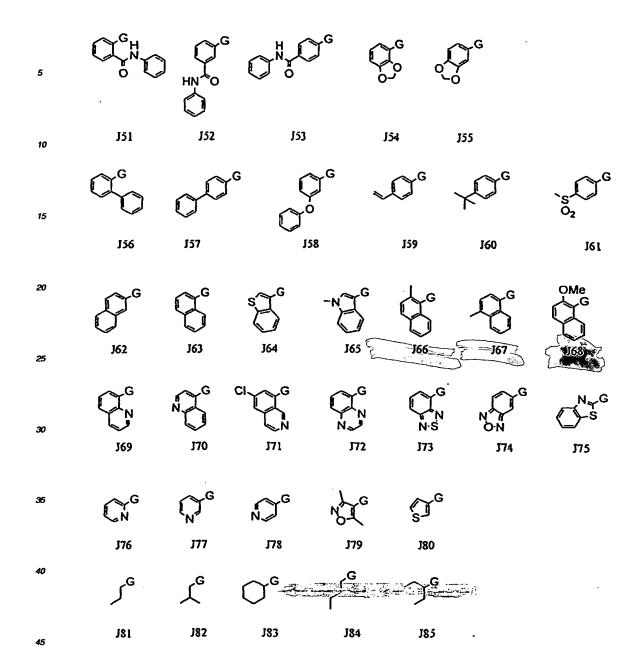


Table 1

50

Compound No.	R ¹	R ²	Α	E	G	J	Э	Х
1	Ι	H	A 1	СООН	CH ₂ CH ₂	J1	0	СН
2	Н	Н	A1	соон	CH ₂	J2	0	СН
3	Н	I	A 1	СООН	CH ₂	J3	0	СН
4	Н	Н	A 1	СООН	CH ₂	J4	0	СН

Table 1 (continued)

5

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
5	Н	Н	A1	соон	CH ₂	J5	0	СН
6	Н	Н	A1	соон	CH ₂	J6	0	СН
7	Н	Н	A1	соон	CH ₂	J7	0	СН
8	Η	Н	A1	СООН	CH ₂	J8	0	СН
9	Н	Η	A 1	СООН	CH ₂	J9	0	СН
10	Ι	Н	A1	СООН	CH ₂	J10	0	СН
11	Н	H	A 1	СООН	CH ₂	J11	0	СН
12	Ι	Ι	A1	СООН	CH ₂	J12	0	СН
13	Ξ	Ξ	A1	СООН	CH ₂	J13	0	СН
14	Η	Ξ	A1	СООН	CH ₂	J14	٥	СН
15	Н	Ŧ	A1	СООН	CH ₂	J15	0	СН
16	H	Η	A1	СООН	CH ₂	J16	0	СН
17	Н	Н	A 1	СООН	CH ₂	J17	0	СН
18	Н	H	A 1	СООН	CH ₂	J18	0	СН
19	Н	Н	A 1	СООН	CH ₂	J19	0	СН
20	Н	I	A1	СООН	CH ₂	J20	0	СН
21	Н	Н	A 1	СООН	CH ₂	J21	0	СН
22	Н	Н	A1	СООН	CH ₂	J22	0	СН
23	Н	Н	A1	соон	CH ₂	J23	0	CH
24	Н	Н	A 1	СООН	CH ₂	J24	0	СН
25	Н	Н	A 1	СООН	CH ₂	J25	0	СН

Table 2

Compound No.	R ¹	R ²	Α	Е	G	J	m	Х
26		H	_A1_	СООН	CH ₂	J26	0	СН
27	H	" ² H"	À1	СООН	CH2	J27	0	СН
28	Ι	Η	A1	СООН	CH ₂	J28	0	СН
29	Н	Ι	A1	СООН	CH ₂	J29	0	СН
30	Η	I	A1	соон	CH ₂	J30	0	СН
31	Ι	Ι	A 1	СООН	CH ₂	J31	0	СН
32	Ξ	Η	A1	СООН	CH ₂	J32	0	СН
33	Ħ	Ι	A 1	СООН	CH ₂	J33	0	СН
34	H	Ŧ	A 1	СООН	CH ₂	J34	0	СН
35	H	Η	A1	соон	CH ₂	J35	0	СН
36	H	Н	A 1	СООН	CH ₂	J36	0	СН
37	Н	Н	A 1	соон	CH ₂	J37	0	СН
38	Н	Н	A 1	СООН	CH ₂	J38	0	СН

Table 2 (continued)

5

Compound No.	R ¹	R ²	Α	Ε	G	J	m	Х
39	Н	Н	A1	соон	CH ₂	J 39	0	СН
40	Н	Н	A1	соон	CH ₂	J40	0	СН
41	Н	Н	A1	СООН	CH ₂	J41	0	СН
42	Н	Н	A1	СООН	CH ₂	J42	0	СН
43	Н	Н	A1	СООН	CH ₂	J43	0	СН
44	Н	Н	A 1	соон	CH ₂	J44	0	СН
45	Н	Н	A1	СООН	CH ₂	J45	0	CH
46	H	Н	A1	соон	CH ₂	J46	0	СН
47	Н	Н	A1	СООН	CH ₂	J47	0	СН
48	Ħ	Н	A1	соон	CH ₂	J48	0	СН
49	Н	Н	A1	СООН	CH ₂	J49	0	СН
50	Н	Н	A 1	СООН	CH ₂	J50	0	СН

Compound No.	R ¹	R ²	Α	Е	G	J	m	Х
51	Н	Н	A 1	СООН	CH ₂	J51	0	СН
52	Н	Н	A1	СООН	CH ₂	J52	0	СН
53	Н	Н	A 1	СООН	CH ₂	J53	0	СН
54	Н	Н	A1	соон	CH ₂	J54	0	СН
55	Н	Н	A 1	соон	CH ₂	J55	0	СН
56	Н	Н	A 1	COOH	CH ₂	J56	0	СН
57	Η	H	A 1	соон	CH ₂	J57	0	СН
58	Н	Η	A 1	СООН	CH ₂	J58	0	СН
59	Τ	Ŧ	A1	СООН	CH ₂	J59	0	СН
60	, H.,	H SWC 16	A1	СООН	CH ₂	J60	0	СН
61	H	H	Ä1	СООН	CH ₂	J61	0	СН
62	Н	Н	A1	СООН	CH ₂	J62	0	СН
63	Н	Н	A 1	СООН	CH ₂	J63	0	СН
64	H	Н	A1	СООН	CH ₂	J64	0	СН
65	Н	Н	A1	соон	CH ₂	J65	0	СН
66	Н	Н	A1	СООН	CH ₂	J66	0	СН
67	Н	Н	A1	СООН	CH ₂	J67	0	СН
68	Ι	Η	A1	СООН	CH ₂	J68	0	СН
69	Ξ	H	A1	COOH	CH ₂	J69	0	CH
70	Η	H	A1	СООН	CH ₂	J70	0	CH
71	Н	Н	A1	СООН	CH ₂	J71	0	СН
72	Н	Н	A1	СООН	CH ₂	J72	0	СН

Table 3 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
73	Н	Н	A1	соон	CH ₂	J73	0	СН
74	Н	H	A1	СООН	CH ₂	J74	0	СН
75	Н	H	A1	СООН	CH ₂	J75	0	СН

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Table 4

Compound No.	R ¹	R ²	Α	Е	G	J	m	Х
76	Н	Н	A1	соон	CH ₂	J76	0	СН
77	Н	Н	A1	соон	CH ₂	J77	0	СН
78	Н	Н	A1	СООН	CH ₂	J78	0	СН
79	Н	Н	A1	СООН	CH ₂	J79	0	СН
80	Н	Н	A1	СООН	CH ₂	J80	0	СН
81	Me	Me	A1	СООН	CH ₂	J1	0	СН
82	Ме	Me.	A1	СООН	CH ₂	J2	0	СН
83	Ме	Me	A1	СООН	CH ₂	J3	0	СН
84	Me	Me	A1	СООН	CH ₂	J4	0	СН
85	Me	Ме	A 1	СООН	CH ₂	J5	0	СН
86	Ме	Ме	A1	СООН	CH ₂	J6	0	СН
87	Me	Ме	A1	СООН	CH ₂	J7	0	СН
88	Me	Ме	A1	СООН	CH ₂	J8	0	СН
89	Me	Me	A 1	СООН	CH ₂	J9	0	СН
90	Me	Me	A 1	СООН	CH ₂	J10	0	СН
91	Me	Ме	A1	СООН	CH ₂	J11	0	СН
92	Me	Ме	A 1	СООН	CH ₂	J12	0	СН
93	Me	Ме	A1	СООН	CH ₂	J13	0	СН
94	Me	"Me	A1_	СООН	CH ₂	J14	0	СН
95	Me	Ме	A1	"СООН	CH ₂	J15	0	СН
96	Me	Ме	A1	СООН	CH ₂	J16	0	СН
97	Ме	Me	A 1	СООН	CH ₂	J17	0	СН
98	Me	Ме	A 1	СООН	CH ₂	J18	0	СН
99	Ме	Me	A 1	СООН	CH ₂	J19	0	СН
100	Me	Me	A1	СООН	CH ₂	J20	0	СН

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Compound No.	R ¹	R ²	Α	E	G	٦	3	Х
101	Ме	Me	A 1	СООН	CH ₂	J21	0	СН

Table 5 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	X
102	Me	Me	A1	соон	CH ₂	J22	0	СН
103	Ме	Ме	A1	соон	CH ₂	J23	0	СН
104	Ме	Me	A1	соон	CH ₂	J24	0	СН
105	Me	Ме	A1	соон	CH ₂	J25	0	СН
106	Ме	Me	A1	соон	CH ₂	J26	0	СН
107	Me	Ме	A 1	соон	CH ₂	J27	0	СН
108	Ме	Me	A1	соон	CH ₂	J28	0	СН
109	Me	Ме	A1	соон	CH ₂	J29	0	СН
110	Ме	Ме	A1	соон	CH ₂	J30	0	СН
111	Me	Ме	A1	соон	CH ₂	J31	0	СН
112	Ме	Ме	A1	СООН	CH ₂	J32	0	СН
113	Me	Me	A1	соон	CH ₂	J33	0	СН
114	Me	Ме	A1	СООН	CH ₂	J34	0	СН
115	Me	Ме	A1	соон	CH ₂	J35	0	СН
116	Ме	Ме	A1	соон	CH ₂	J36	0	СН
117	Ме	Me	A1	соон	CH ₂	J37	0	СН
118	Me	Me	A 1	СООН	CH ₂	J38	0	СН
119	Me	Ме	A 1	СООН	CH ₂	J39	0	СН
120	Me	Me	A1	СООН	CH ₂	J40	0	СН
121	Me	Ме	A 1	соон	CH ₂	J41	0	СН
122	Ме	Me	A 1	соон	CH ₂	J42	0	СН
123	Ме	Ме	A 1	соон	CH ₂	J43	0	СН
124	Ме	Me	A1	COOH	CH ₂	J44	0	СН
125	Me	Ме	A1	СООН	CH ₂	J45	0	СН

Table 6

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
126	Ме	Ме	A1	соон	CH ₂	J46	0	СН
127	Ме	Ме	A1	СООН	CH ₂	J47	0	СН
128	Me	Ме	A1	соон	CH ₂	J48	0	СН
129	Me	Ме	A1	СООН	CH ₂	J49	0	СН
130	Ме	Ме	A1	соон	CH ₂	J50	0	СН
131	Ме	Ме	A1	соон	CH ₂	J51	0	СН
132	Me	Me	A1	СООН	CH ₂	J52	0	CH
133	Me	Me	A 1	соон	CH ₂	J53	0	СН
134	Ме	Ме	A 1	СООН	CH ₂	J54	0	СН

Table 6 (continued)

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Compound No.	R ¹	R ²	Α	E	G	J	m	Х
135	Me	Me	A1	СООН	CH ₂	J55	0	СН
136	Ме	Ме	A1	СООН	CH ₂	J56	0	СН
137	Me	Me	A1	СООН	CH ₂	J57	0	СН
138	Me	Me	A 1	СООН	CH ₂	J58	0	СН
139	Ме	Ме	A1	СООН	CH ₂	J59	0	СН
140	Ме	Ме	A 1	СООН	CH ₂	J60	0	СН
141	Ме	Me	A 1	COOH	CH ₂	J61	0	СН
142	Ме	Me	A 1	COOH	CH ₂	J62	0	СН
143	Ме	Ме	A 1	СООН	CH ₂	J63	0	СН
144	Ме	Me	A1	СООН	CH ₂	J64	0	СН
145	Ме	Me	A1	СООН	CH ₂	J65	0	СН
146	Ме	Me	A1	СООН	CH ₂	J66	0	СН
147	Ме	Me	A 1	СООН	CH ₂	J67	0	СН
148	Me	Ме	A 1	СООН	CH ₂	J68	0	СН
149	Me	Me	A 1	СООН	CH ₂	J69	0	СН
150	Ме	Ме	A 1	СООН	CH ₂	J70	0	СН

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Compound No.	R ¹	R ²	Α	E	G	J	Э	Х
151	Ме	Ме	A 1	СООН	CH ₂	J71	0	СН
152	Me	Ме	A1	СООН	CH ₂	J72	0	CH
153	Ме	Ме	A 1	СООН	CH ₂	J73	0	СН
154	Ме	Ме	A1	СООН	CH ₂	J74	0	СН
155	Ме	Ме	A1	СООН	CH ₂	J75	0	СН
156	Me	_Me	A1-,	СООН	CH ₂	J76	0	СН
157	Ме	Ме	A1	СООН	CH ₂	J77	0	СН
158	Ме	Me	A1	СООН	CH ₂	J78	0	СН
159	Ме	Ме	A1	соон	CH ₂	J79	0	СН
160	Ме	Ме	A1	соон	CH ₂	J80	0	СН
161	CI	CI	A1	СООН	CH ₂ CH ₂	J1	0	СН
162	CI	CI	A1	СООН	CH ₂	J4	0	СН
163	CI	CI	A1	СООН	CH ₂	J10	0	СН
164	CI -	CI	A1	СООН	CH ₂	J18	0	СН
165	CI	CI	A1	СООН	CH ₂	J21	0	СН
166	CI	СІ	A1	СООН	CH ₂	J28	0	СН
167	CI	CI	A1	СООН	CH ₂	J35	0	СН

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Table 7 (continued)

Compound No.	R ¹	R ²	Α	Ε	G	J	m	Х
168	CI	CI	A1	СООН	CH ₂	J37	0	СН
169	CI	CI	A1	СООН	CH ₂	J39	0	СН
170	CI	CI	A1	СООН	CH ₂	J43	0	СН
171	CI	CI	A1	СООН	CH ₂	J46	0	СН
172	CI	CI	A1	СООН	CH ₂	J50	0	СН
173	CI	CI	A1	СООН	CH ₂	J54	0	СН
174	CI	CI	A1	СООН	CH ₂	J63	0	СН
175	CI	CI	A1	СООН	CH ₂	J64	0	СН

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Table 8

Compound No.	R ¹	R ²	Α	Ε	G	J	m	X
176	CI	CI	A1	СООН	CH ₂	J65	0	СН
177	CI	CI	A1	СООН	CH ₂	J66	0	СН
178	CI	Cl	A1	соон	CH ₂	J67	0	СН
179	CI	CI	A1	СООН	CH ₂	J71	0	СН
180	-CH ₂ Cl	H ₂ CH ₂ -	A1	СООН	CH ₂ CH ₂	J1	0	СН
181	-CH ₂ Cl	H ₂ CH ₂ -	A1	СООН	CH ₂	J4	0	СН
182	-CH ₂ Cl	H ₂ CH ₂ -	A1	СООН	CH ₂	J10	0	СН
183	-CH ₂ Cl	H ₂ CH ₂ -	A1	соон	CH ₂	J18	0	СН
184	-CH ₂ Cl	H ₂ CH ₂ -	A1	СООН	CH ₂	J21	0	СН
185	-CH ₂ Cl	H ₂ CH ₂ -	A1	соон	CH ₂	J28	0	СН
186	-CH ₂ Cl	H ₂ CH ₂ -	A1	соон	CH ₂	J 35	0	СН
187	-CH ₂ Cl	H ₂ CH ₂ -	A1	СООН	CH ₂	J37	0	СН
188	-CH ₂ Cl	H ₂ CH ₂ -	A1	СООН	CH ₂	J39	0	СН
1-129 - J-4	GH ₂ C	t ₂ CH ₂ -&	gA1	СООН	CH ₂	J43	0	СН
190	-CH ₂ C	H ₂ CH ₂ -	A1	соон	CH ₂	J46	0	СН
191	-CH ₂ Cl	H ₂ CH ₂ -	A1	соон	CH ₂	J50	0	СН
192	-CH ₂ C	H ₂ CH ₂ -	A1	соон	CH ₂	J54	0	СН
193	-CH ₂ C	H ₂ CH ₂ -	A1	соон	CH ₂	J63	0.	СН
194	-CH ₂ Cl	H ₂ CH ₂ -	A1	соон	CH ₂	J64	0	СН
195	-CH ₂ C	H ₂ CH ₂ -	A1	соон	CH ₂	J65	0	СН
196	-CH ₂ C	H ₂ CH ₂ -	A1	соон	CH ₂	J66	0	СН
197	-CH ₂ C	H ₂ CH ₂ -	A1	соон	CH ₂	J67	0	СН
198	-CH ₂ Cl	H ₂ CH ₂ -	A1	соон	CH ₂	J71	0	СН
199	-OCH ₂ O-		A1	соон	CH ₂ CH ₂	J1	0	СН
200	-oc	H ₂ O-	A1	соон	CH ₂	J4	0	СН

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Table 9

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	Compound No.	R ¹	R ²	Α	E	G	J	m	Х
	201	-oc	H ₂ O-	A1	соон	CH ₂	J10	0	СН
	202	-oc	H ₂ O-	A1	соон	CH ₂	J18	0	СН
	203	-OCI	H ₂ O-	A1	соон	CH ₂	J21	0	СН
	204	-OC	H ₂ O-	A1	соон	CH ₂	J28	0	СН
	205	-OCI	H ₂ O-	A1	соон	CH ₂	J35	0	СН
	206	-oci	1 ₂ O-	A1	соон	CH ₂	J37	0	СН
	207	-OCI	1 ₂ O-	A1	соон	CH ₂	J39	0	СН
	208	-OCI	H ₂ O-	A1	СООН	CH ₂	J43	0	СН
	209	-OCI	H ₂ O-	A1	соон	CH ₂	J46	0	СН
	210	-OCI	H ₂ O-	A1	соон	CH ₂	J50	0	СН
	211	-OCI	H ₂ O-	A1	СООН	CH ₂	J54	0	СН
	212	-OCI	1 ₂ O-	A1	соон	CH ₂	J63	0	СН
	213	-OCI	1 ₂ O-	A1	СООН	CH ₂	J64	0	СН
	214	-OCI	1 ₂ O-	A 1	соон	CH ₂	J65	0	СН
	215	-OCł	1 ₂ O-	A1	СООН	CH ₂	J66	0	СН
	216	-OCI	1 ₂ O-	A 1	соон	CH ₂	J67	0	СН
	217	-OCI	1 ₂ O-	A1	СООН	CH ₂	J71	0	СН
	218	-OCH ₂ (CH ₂ O-	A 1	соон	CH ₂ CH ₂	J1	0	СН
,	219	-OCH ₂	CH ₂ O-	A1	COOH	CH ₂	J4	0	СН
	220	-OCH ₂	CH ₂ O-	A1	COOH	CH ₂	J10	0	CH
	221	-OCH ₂ (CH ₂ O-	A1	СООН	CH ₂	J18	0	СН
	222	-OCH ₂ (CH ₂ O-	A1	СООН	CH ₂	J35	0	СН
	223	-OCH ₂ 6	CH ₂ O-	A 1	СООН	CH ₂	J37	0	СН
	224	-OCH ₂ 6	CH ₂ O-	A 1	соон	CH ₂	J39	0	СН
,		OCH	CH ₂ O-	A1	COÓH	CH ₂	J50	0	СН
-									

Compound No.	R ¹	R ²	A	E	G	J	m	Х
226	-OCH ₂	CH ₂ O-	A1	соон	CH ₂	J63	0	СН
227	-OCH ₂	CH ₂ O-	A1	СООН	CH ₂	J64	0	СН
228	-OCH ₂	CH ₂ O-	A1	соон	CH ₂	J65	0	СН
229	-OCH ₂	CH ₂ O-	A1	соон	CH ₂	J67	0	СН
230	-OCH ₂ CH ₂ O-		A1	соон	CH ₂	J71	0	СН
231	OMe	OMe	A1	соон	CH ₂ CH ₂	J1	0	СН

Table 10 (continued)

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Compound No.	R ¹	R ²	Α	E	G	J	m	Х
232	OMe	OMe	A1	СООН	CH ₂	J4	0	СН
233	OMe	OMe	A1	СООН	CH ₂	J10	0	СН
234	OMe	OMe	A1	СООН	CH ₂	J18	0	СН
235	OMe	OMe	A1	соон	CH ₂	J35	0	СН
236	OMe	OMe	A1	СООН	CH ₂	J37	0	СН
237	OMe	OMe	A1	СООН	CH ₂	J39	0	СН
238	OMe	OMe	A1	СООН	CH ₂	J50	0	СН
239	OMe	OMe	A1	СООН	CH ₂	J63	0	СН
240	OMe	OMe	A1	СООН	CH ₂	J64	0	СН
241	OMe	OMe	A1	СООН	CH ₂	J65	0	СН
242	OMe	OMe	A1	СООН	CH ₂	J67	0	СН
243	OMe	OMe	A 1	COOH	CH ₂	J71	0	СН
244	F	F	A1	СООН	CH ₂	J35	0	СН
245	F	F	A1	COOH	CH ₂	J37	0	СН
246	F	F	A1	СООН	CH ₂	J39	0	СН
247	F	F	A 1	СООН	CH ₂	J50	0	СН
248	F	F	A1	СООН	CH ₂	J63	0	СН
249	F	F	A1	СООН	CH ₂	J64	0	СН
250	F	F	A1	COOH	CH ₂	J65	0	СН

Table 11

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	Compound No.	R ¹	R ²	Α	E	G	J	m	Х
	251	F	F	A1	соон	CH ₂	J67	0	CH
	252	Н	Н	A1	соон	CH ₂	J35	0	N
i i	253	H	Н	A 1	СООН	CH ₂	J37	0	N
i i Tel	254	Ήr	Н	A1	соон	CH ₂	J39	0	N
	255	Н	Н	A 1	СООН	CH ₂	J50	0	N
	256	Н	Н	A1	СООН	CH ₂	J63	0	N
	257	Н	Н	A1	СООН	CH ₂	J64	0	N
	258	Н	Н	A 1	соон	CH ₂	J65	0	N
	259	Н	Н	A 1	соон	CH ₂	J67	0	N
	260	Me	Н	A1	соон	CH ₂	J35	0	СН
•	261	Me	Н	A 1	СООН	CH ₂	J37	0	СН
	262	Ме	Н	A1	СООН	CH ₂	J39	0	СН
	263	Ме	Н	A 1	СООН	CH ₂	J50	0	СН
	264	Me	Н	A 1	соон	CH ₂	J63	0	СН
	265	Me	H	A 1	СООН	CH ₂	J64	0	СН

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Table 11 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
266	Me	Н	A 1	соон	CH ₂	J65	0	СН
267	Me	Н	A1	СООН	CH ₂	J67	0	СН
268	OMe	Н	A1	соон	CH ₂	J35	0	СН
269	OMe	Н	A1	СООН	CH ₂	J37	0	СН
270	OMe	Н	A1	СООН	CH ₂	J39	0	СН
271	OMe	Н	A 1	СООН	CH ₂	J50	0	СН
272	OMe	Н	A1	СООН	CH ₂	J63	0	СН
273	OMe	Н	A 1	СООН	CH ₂	J64	0	СН
274	ОМе	Ι	A 1	СООН	CH ₂	J65	0	СН
275	OMe	Ξ	A 1	СООН	CH ₂	J67	0	СН

Table 12

			14	DIE 12				
Compound No.	R ¹	R ²	Α	E	G	J	Э	Х
276	OEt	Н	A1	СООН	CH ₂	J63	0	СН
277	OEt	Н	A1	соон	CH ₂	J64	0	СН
278	OEt	н	A1	соон	CH ₂	J65	0	СН
279	CF3	Н	A1	соон	CH ₂	J63	0	СН
280	CF3	Н	A1	соон	CH ₂	J64	0	СН
281	CF3	Н	A1	соон	CH ₂	J65	0	СН
282	CN	Н	A1	СООН	CH ₂	J63	0	СН
283	CN	Н	A1	соон	CH ₂	J64	٥	СН
284	CN	Н	A1	СООН	CH ₂	J65	0	СН
285	CI	Н	A1	соон	CH ₂	J63	0	N
286	CI	. H	A1	соон	CH ₂	J64	0	N
287	CI	Н	A1	СООН	CH ₂	J65	0	N
288	Me	Ме	A2	СООН	CH ₂	J35	0	СН
289	Me	Me	A 2	СООН	CH ₂	J37	0	СН
290	Me	Ме	A2	СООН	CH ₂	J39	0	СН
291	Me	Me	A2	СООН	CH ₂	J63	0	СН
292	Me	Ме	A2	соон	CH ₂	J64	0	СН
293	Ме	Ме	A2	соон	CH ₂	J65	0	СН
294	Me	Ме	A2	соон	CH ₂ CH ₂	J1	0	СН
295	Me	Ме	А3	соон	CH ₂	J1	0	СН
296	Me	Ме	А3	соон	CH ₂	J35	0	СН
297	Me	Me	A3	соон	CH ₂	J37	0	СН
298	Me	Ме	A3	соон	CH ₂	J39	0	СН
299	Me	Ме	A3	соон	CH ₂	J50	0	СН

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Table 12 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
300	Me	Ме	А3	СООН	CH ₂	J63	0	CH

Table 13

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Compound No.

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Table	1	4
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Compound No.	R ¹	R ²	Α	E	G	J	m	×
326	CI	CI	A4	соон	CH ₂	J65	0	СН
327	Н	Н	A4	СООН	CH ₂	J37	0	N
328	Η	Н	A4	СООН	CH ₂	J39	0	N

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Table 14 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
329	Н	Н	A4	соон	CH ₂	J63	0	N
330	Н	Н	A4	соон	CH ₂	J64	0	N
331	Н	н	A4	соон	CH ₂	J65	0	N
332	Me	Ме	A5	соон	CH ₂	J1	0	СН
333	Me	Ме	A 5	соон	CH ₂ CH ₂	J1	0	СН
334	Me	Me	A6	СООН	CH ₂	J1	0	СН
335	Ме	Me	A6	соон	CH ₂ CH ₂	J1	0	СН
336	Me	Ме	A7	СООН	CH ₂	J1	0	СН
337	Me	Ме	A7	СООН	CH ₂ CH ₂	J1	0	СН
338	Ме	Me	A8	СООН	CH ₂	J1	0	СН
339	Me	Me	A8	СООН	CH ₂ CH ₂	J1	0	СН
340	Me	Me	A 9	СООН	CH ₂	J1	0	СН
341	Me	Ме	A 9	СООН	CH ₂ CH ₂	J1	0	СН
342	Me	Ме	A10	соон	CH ₂	J1	0	СН
343	Me	Ме	A10	соон	CH ₂ CH ₂	J1	0	СН
344	Me	Me	A11	соон	CH ₂	J37	0	СН
345	Me	Me	A11	соон	CH ₂	J39	0	СН
346	Me	Ме	A11	соон	CH ₂	J50	0	СН
347	Me	Ме	A11	соон	CH ₂	J63	0	СН
348	Me	Ме	A11	соон	CH ₂	J64	0	СН
349	Н	Н	A11	соон	CH ₂	J37	0	СН
350	Н	Н	A11	соон	CH ₂	J 39	0	СН

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Table 15

Compound No. R¹ R2 Α G Х J m 351 Н Н A11 СООЙ CH₂ J50 0 СН 352 Н Н A11 COOH CH₂ СН **J63** 0 353 Н Н A11 COOH CH₂ **J64** 0 СН 354 Н Н A11 COOH J65 СН CH₂ 0 355 CI CI A11 COOH CH₂ **J37** СН 0 356 CI CI A11 COOH CH₂ **J**39 СН 0 357 CI CI A11 COOH CH₂ J50 0 CH 358 CI CI A11 COOH CH₂ **J63** CH 0 359 CI CI A11 СООН CH₂ **J64** СН 0 360 CI СООН CI A11 CH₂ J65 0 CH 361 Н Н A11 СООН CH₂ **J37** N

Table 15 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
362	Н	Н	A11	соон	CH ₂	J 39	0	N
363	Н	Н	A11	соон	CH ₂	J50	0	N
364	Н	Н	A11	соон	CH ₂	J63	0	N
365	Н	Н	A11	соон	CH ₂	J64	0	Ŋ
366	Н	Н	A11	соон	CH ₂	J65	0	Z
367	Ме	Me	A12	СООН	CH ₂	J1	0	СН
368	Ме	Me	A12	СООН	CH ₂ CH ₂	J1	0	CH
369	Ме	Me	A13	СООН	CH ₂	J1	0	СН
370	Ме	Me	A13	СООН	CH ₂ CH ₂	J1	0	СН
371	Ме	Me	A14	СООН	CH ₂	J1	0	СН
372	Ме	Me	A14	СООН	CH ₂ CH ₂	J1	0	СН
373	Ме	Me	A15	СООН	CH ₂	J1	0	СН
374	Me	Me	A15	COOH	CH ₂ CH ₂	J1	0	СН
375	Me	Ме	A16	соон	CH ₂	J1	0	СН

Table 16

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
376	Me	Me	A16	СООН	CH ₂ CH ₂	J1	0	СН
377	Ме	Me	A16	СООН	CH ₂	J37	0	СН
378	Me	Ме	A16	СООН	CH ₂	J39	0	СН
379	Me	Me	A16	соон	CH ₂	J50	0	СН
380	Me	Me	A16	СООН	CH ₂	J63	0	СН
381	Me	Me	A16	СООН	CH ₂	J64	0	СН
382	Me	Me	A16	СООН	CH ₂	J65	0	CH
383	H	H	A16	СООН	CH ₂	J37	0	끉
384	Н	Н	A16	СООН	CH ₂	739	0	СН
385	Ι	Ξ	A16	СООН	CH ₂	J 50	0	СН
386	Н	Ξ	A16	СООН	CH ₂	J63	0	СН
387	Н	Н	A16	COOH	CH ₂	J64	0	CH
388	Н	Н	A16	СООН	CH ₂	J65	0	СН
389	Ме	Ме	A17	СООН	CH ₂	J1	0	СН
390	Ме	Ме	A17	соон	CH ₂ CH ₂	J1	0	СН
391	Ме	Ме	A18	соон	CH ₂ CH ₂	J1	0	СН
392	Ме	Ме	A18	соон	CH ₂	J37	0	СН
393	Ме	Me	A18	СООН	CH ₂	J39	0	СН
394	Ме	Ме	A18	соон	CH ₂	J50	0	СН
395	Ме	Ме	A18	соон	CH ₂	J63	0	СН

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Table 16 (continued)

Compound No.	R ¹	R ²	A	E	G	J	m	Х
396	Me	Me	A18	соон	CH ₂	J64	0	СН
397	Me	Me	A18	соон	CH ₂	J65	0	СН
398	Н	Н	A18	соон	CH ₂	J37	0	СН
399	Н	Н	A18	соон	CH ₂	J39	0	СН
400	Н	Н	A18	соон	CH ₂	J50	0	СН

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			Table						_
Compound No.	R ¹	R ²	Α	E	G	J	m	Х	
401	Н	Н	A18	соон	CH ₂	J63	0	СН	
402	н	Н	A18	соон	CH ₂	J64	0	СН	
403	Н	Н	A18	соон	CH ₂	J65	0	СН	
404	CI	CI	A18	соон	CH ₂	J37	0	СН	
405	CI	CI	A18	соон	CH ₂	J63	0	СН	
406	CI	CI	A18	СООН	CH ₂	J64	0	СН	
407	CI	CI	A18	СООН	CH ₂	J65	0	СН	
408	Н	Н	A18	соон	CH ₂	J37	0	N	
409	Н	Н	A18	соон	CH ₂	J39	0	N	
410	Н	Н	A18	соон	CH ₂	J63	0	N	
411	Н	Н	A18	СООН	CH ₂	J64	0	N	
412	Н	Н	A18	СООН	CH ₂	J65	0	N	
413	Me	Н	A18	СООН	CH ₂	J37	0	СН	
414	Me	Н	A18	СООН	CH ₂	J39	0	СН	
415	Me	Н	A18	СООН	CH ₂	J63	0	СН	
416	Me	Н	A18	соон	CH ₂	J64	0	СН	
417	Me	Н	A18	СООН	CH ₂	J65	0	CH	
418	OMe	Н	A18	СООН	CH ₂	J37	0	СН	
419	OMe	Н	A18	соон	CH ₂	J39	0	СН	
420	OMe	Н	A18	СООН	CH ₂	J63	0	СН	
421	ОМе	Н	A18	соон	CH ₂	J64	0	СН	1
422	OMe	Н	A18	соон	CH ₂	J65	0	СН	
423	OEt	Н	A18	соон	CH ₂	J63	0	СН	
424	OEt	Н	A18	соон	CH ₂	J64	0	СН	
425	OEt	Н	A18	СООН	CH ₂	J65	0	СН	

Table 18

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			Ial	He 18						
Compound No.	R ¹	R ²	Α	E	G	J	m	х		
426	CF3	Н	A18	СООН	CH ₂	J63	0	СН		
427	CF3	н	A18	соон	CH ₂	J64	0	СН		
428	CF3	Н	A18	СООН	CH ₂	J65	0	СН		
429	CN	Н	A18	СООН	CH ₂	J63	0	СН		
430	CN	Н	A18	СООН	CH ₂	J64	0	СН		
431	CN	Н	A18	соон	CH ₂	J65	0	СН		
432	F	Н	A18	соон	CH ₂	J63	0	СН		
433	F	Н	A18	COOH	CH ₂	J64	0	СН		
434	F	Н	A18	СООН	CH ₂	J65	0	СН		
435	CI	Н	A18	СООН	CH ₂	J63	0	N		
436	CI	Н	A18	COOH	CH ₂	J64	0	N		
437	CI	Н	A18	СООН	CH ₂	J65	0	N		
438	Н	н	A18	СООН	CH ₂	J37	0	N		
439	Me	Ме	A19	СООН	CH ₂	J1	0	СН		
440	Ме	Ме	A19	COOH	CH ₂ CH ₂	J1	0	СН		
441	Me	Ме	A19	СООН	CH ₂	J37	0	СН		
442	Ме	Ме	A19	соон	CH ₂	J39	0	СН		
443	Ме	Ме	A19	СООН	CH ₂	J50	0	СН		
444	Ме	Ме	A19	СООН	CH ₂	J63	0	СН		
445	Me	Ме	A19	СООН	CH ₂	J64	0	СН		
446	Me	Me	A19`	СООН	CH ₂	J65	0	СН		
447	Н	Н	A19	СООН	CH ₂	J1	0	СН		
448	Н	Н	A19	COOH	CH ₂ CH ₂	J1	0	СН		
449	Н	Н	A19	СООН	CH ₂	J37	0	СН		
450	Н	Н	A19	СООН	CH ₂	J39	0	СН		
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Table 19

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
451	Н	Н	A19	СООН	CH ₂	J50	0	СН
452	Н	Н	A19	СООН	CH ₂	J63	0	СН
453	Н	Н	A19	СООН	CH ₂	J64	0	СН
454 /	Н	Н	A19	СООН	CH ₂	J65	0	СН
XEE/	Ме	Ме	A20	СООН	CH ₂	J64	0	СН
A56	Ме	Ме	A20	СООН	CH ₂	J65	0	СН
457	Ме	Ме	A20	СООН	CH ₂	J67	0	СН

Table 19 (continued)

	-1	-2						1
Compound No.	R ¹	R ²	Α	Ε	G	J	æ	X
458	Me	Me	A20	СООН	CH ₂	J71	0	СН
459	Н	Н	A20	СООН	CH ₂	J64	0	СН
460	Н	Н	A20	СООН	CH ₂	J65	0	СН
461	Н	Н	A20	СООН	CH ₂	J67	0	СН
462	Н	Н	A20	СООН	CH ₂	J71	0	СН
463	CI	CI	A20	СООН	CH ₂	J64	0	СН
464	CI	CI	A20	СООН	CH ₂	J65	0	СН
465	CI	СІ	A20	СООН	CH ₂	J67	0	СН
466	CI	CI	A20	СООН	CH ₂	J71	0	СН
467	Н	Н	A20	СООН	CH ₂	J64	0	N
468	Н	Н	A20	СООН	CH ₂	J65	0	N
469	н	Н	A20	СООН	CH ₂	J67	0	N
470	Н	Н	A20	СООН	CH ₂	J71	0	N
471	Ме	Н	A20	СООН	CH ₂	J64	0	СН
472	Me	Н	A20	СООН	CH ₂	J65	0	СН
473	Ме	Н	A20	СООН	CH ₂	J67	0	СН
474	Me	Н	A20	соон	CH ₂	J71	0	СН
475	OMe	Н	A20	соон	CH ₂	J64	0	СН

Table 20

			Iable	20					
Compound No.	R ¹	R ²	Α	E	G	J	m	Х	
476	OMe	Н	A20	СООН	CH ₂	J65	0	СН	
477	OMe	Н	A20	СООН	CH ₂	J67	0	СН	
478	OMe	Н	A20	соон	CH ₂	J71	0	СН	
479	OEt	Н	A20	соон	CH ₂	J64	Q.	: 아.	
480	OEt	Н	A20	СООН	CH ₂	J65	0	СН	and an analysis of the second
481	OEt	Н	A20	СООН	CH ₂	J67	0	СН	
482	OEt	Н	A20	СООН	CH ₂	J71	0	СН	
483	F	Н	A20	СООН	CH ₂	J64	0	CH	
484	F	Н	A20	СООН	CH ₂	J65	0	СН	
485	F	Н	A20	соон	CH ₂	J67	0	СН	
486	F	Н	A20	соон	CH ₂	J71	0	СН	
487	CF3	Н	A20	соон	CH ₂	J64	0	СН	
488	CF3	Н	A20	СООН	CH ₂	J65	0	СН	
489	CF3	Н	A20	СООН	CH ₂	J67	0	СН	
490	CF3	Н	A20	соон	CH ₂	J71	0	СН]

Table 20 (continued)

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Compound No.	R ¹	R ²	Α	E	G	J	E	Х
491	CN	Н	A20	СООН	CH ₂	J64	0	СН
492	CN	Н	A20	соон	CH ₂	J65	0	СН
493	CN	Н	A20	соон	CH ₂	J67	0	СН
494	CN	Н	A20	СООН	CH ₂	J71	0	СН
495	CI	Н	A20	СООН	CH ₂	J64	0	N
496	CI	Н	A20	СООН	CH ₂	J 6 5	0	N
497	CI	Н	A20	соон	CH ₂	J67	0	N
498	CI	Н	A20	соон	CH ₂	J71	0	N
499	Н	Н	A21	соон	CH ₂	J63	0	СН
500	Н	Н	A21	COOH	CH ₂	J65	0	СН

Table 21

Compound No.	R ¹	R ²	Α	E	G	J	æ	Х
501	Me	Me	A1	соон	CH ₂ CH ₂	J1	0	СН
502	Ме	Me	A1	СООН	CH ₂ CH ₂	J37	0	СН
503	Ме	Me	A 1	СООН	CH ₂ CH ₂	J39	0	СН
504	Me	Me	A 1	СООН	CH ₂ CH ₂	J50	0	СН
505	Me	Me	A 1	СООН	CH ₂ CH ₂	J62	0	СН
506	Me	Me	A1	СООН	CH ₂ CH ₂	J63	0	СН
507	Ме	Me	A1	СООН	CH ₂ CH ₂	J64	0	СН
508	Ме	Ме	A1	СООН	CH ₂ CH ₂	J65	0	СН
509	н	Н	A1	соон	CH ₂ CH ₂	J1	0	СН
510	Н	Н	A1	соон	CH ₂ CH ₂	J37	0	СН
511	Н	H	A1	СООН	CH ₂ CH ₂	J39	0	СН
512	н	Н	A1	СООН	CH ₂ CH ₂	J50	0	СН
513	н	Н	A1	СООН	CH ₂ CH ₂	J62	10	TCHT.
514	Н	Н	A1	СООН	CH ₂ CH ₂	J63	0	СН
515	Н	Н	A1	СООН	CH ₂ CH ₂	J64	0	СН
516	Н	Н	A1	соон	CH ₂ CH ₂	J65	0	СН
517	Me	Me	A4	СООН	CH ₂ CH ₂	J37	0	СН
518	Ме	Ме	A4	СООН	CH ₂ CH ₂	J39	0	СН
519	Ме	Me	A4	соон	CH ₂ CH ₂	J67	0	СН
520	Me	Me	A4	СООН	CH ₂ CH ₂	J64	0	СН
521	Me	Me	A4	СООН	CH ₂ CH ₂	J65	0	СН
522	Н	Н	A4	соон	CH ₂ CH ₂	J37	0	СН
523	Н	Н	A4	СООН	CH ₂ CH ₂	J39	0	СН
524	н	Н	A4	СООН	CH ₂ CH ₂	J63	0	СН

Table 21 (continued)

Compound No.	R ¹	R ²	Α	Ε	G	J	æ	Х
525	Н	Н	A4	СООН	CH ₂ CH ₂	J64	0	СН

Table 22

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									•
Compound No.	R ¹	R ²	Α	Ε	G	J	m	Х	
526	Н	Н	A 4	СООН	CH ₂ CH ₂	J65	0	СН	
527	Н	Н	A11	соон	CH ₂ CH ₂	J37	0	СН	
528	Н	н	A11	СООН	CH ₂ CH ₂	J39	0	СН	
529	Н	Н	A11	СООН	CH ₂ CH ₂	J63	0	СН	
530	Н	Н	A11	СООН	CH ₂ CH ₂	J64	0	СН	
531	Н	Н	A11	соон	CH ₂ CH ₂	J65	0	СН	
532	Н	Н	A18	соон	CH ₂ CH ₂	J37	0	CH	
533	н	Н	A18	СООН	CH ₂ CH ₂	J39	0	СН	
534	Н	Н	A18	соон	CH ₂ CH ₂	J63	0	СН	
535	Н	Н	A18	СООН	CH ₂ CH ₂	J64	0	СН	
, 536	Н	Н	A18	соон	CH ₂ CH ₂	J65	0	СН	
537	Me	Ме	A20	СООН	CH ₂ CH ₂	J 37	0	СН	
538	Ме	Ме	A20	СООН	CH ₂ CH ₂	J39	0	СН	
539	Me	Ме	A20	соон	CH ₂ CH ₂	J63	0	СН	
540	Ме	Ме	A20	соон	CH ₂ CH ₂	J64	0	СН	
541	Me	Ме	A20	СООН	CH ₂ CH ₂	J65	0	СН	
542	н	Н	A20	соон	CH ₂ CH ₂	J37	0	СН	
543	Н	н	A20	соон	CH ₂ CH ₂	J 39	0	СН	
544	Н	Н	A20	соон	CH ₂ CH ₂	J63	0	СН	
545	Н	Н	A20	соон	CH ₂ CH ₂	J64	0	СН	
546	Н	Н	A20	соон	CH ₂ CH ₂	J65	0	CH	ಎಸ್. ಎಸ್.ಕ್ಷಮ್ಗಳ ಪ್ರಸ್ತಿಸಿಗಳು
547	Me	Me	A1	соон	co	J1==	0	CH *	
548	Ме	Me	A1	СООН	co	J63	0	СН	
549	Н	Н	A1	СООН	co	J1	0	СН	
550	Н	Н	A1	соон	co	J63	0	СН	
									•

Compound No.	R ¹	R ²	Α	E	G	J	E	X
551	Ме	Ме	A4	СООН	co	J1	0	СН
552	Me	Me	A 4	СООН	CO	J63	0	СН
553	Н	Н	A4	СООН	CO	J1	0	СН

Table 23 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	х
554	Н	Н	A4	соон	co	J63	0	СН
555	н	Н	A11	соон	co	J1	0	СН
556	Н	Н	A11	соон	co	J63	0	СН
557	Н	Н	A18	соон	co	J1	0	СН
558	Н	Н	A18	соон	co	J63	0	СН
559	Н	H	A20	соон	co	J1	0	СН
560	Н	Н	A20	соон	со	J63	0	СН
561	Me	Ме	A1	соон	SO ₂	J1	0	СН
562	Me	Me	A1	СООН	SO ₂	J63	0	СН
563	Ŧ	H	A 1	СООН	SO ₂	J1	0	СН
564	Ι	r	A1	СООН	SO ₂	J63	0	СН
565	H	Ι	A4	СООН	SO ₂	J1	0	СН
566	H	Ι	A 4	СООН	SO ₂	J63	0	СН
567	Н	H	A11	СООН	SO ₂	J1	0	СН
568	Н	H	A11	СООН	SO ₂	J63	0	СН
569	Н	H	A18	СООН	SO ₂	J1	0	СН
570	Н	Ξ	A18	СООН	SO ₂	J63	0	СН
571	Н	Н	A20	СООН	SO ₂	J1	0	СН
572	Н	Н	A20	СООН	SO ₂	J63	0	СН
573	Н	Н	A1	СООН	CH ₂ CO	J1	0	СН
574	Н	Н	A1	СООН	CH ₂ CO	J2	0	СН
575	Н	Н	A1	СООН	CH ₂ CO	ß	0	СН

Compound No.	R ¹	R ²	Α	Ε.	G_	- J	<u> </u>	×X.	
576	Н	Н	A1	соон	CH ₂ CO	J4	0	СН	
577	Н	Н	A1	соон	CH ₂ CO	J5	0	СН	
578	Н	Н	A1	соон	CH ₂ CO	J6	0	СН	
579	Н	Н	A1	соон	CH ₂ CO	J7	0	СН	
580	Н	Н	A1	соон	CH ₂ CO	J8	0	СН	
581	Н	Н	A1	соон	CH ₂ CO	J9	0	СН	
582	Н	Н	A1	соон	CH ₂ CO	J10	0	СН	
583	Н	Н	A1	соон	CH ₂ CO	J11	0	СН	
584	Н	Η	A1	соон	CH ₂ CO	J12	0	СН	
585	Н	H	A1	соон	CH ₂ CO	J13	0	СН	
586	Н	Н	A1	СООН	CH ₂ CO	J17	0	СН	

Table 24 (continued)

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Compound No.	R ¹	R ²	Α	E	G	J	m	Х
587	Н	Н	A1	соон	CH ₂ CO	J18	0	СН
588	Н	Н	A1	соон	CH ₂ CO	J19	0	СН
589	Н	Н	A1	соон	CH ₂ CO	J23	0	СН
590	Н	Н	A1	соон	CH ₂ CO	J24	0	СН
591	Н	Н	A1	соон	CH ₂ CO	J25	0	СН
592	Н	H	A1	СООН	CH ₂ CO	J36	0	СН
593	Н	Н	A1	СООН	CH ₂ CO	J47	0	СН
594	Ι	Н	A1	СООН	CH ₂ CO	J57	0	СН
595	Ξ	Н	A 1	СООН	CH ₂ CO	J62	0	СН
596	Ме	Me	A1	СООН	CH ₂ CO	J1	0	СН
597	Me	Me	A 1	СООН	CH ₂ CO	J2	0	СН
598	Me	Me	A 1	СООН	CH ₂ CO	J3	0	СН
599	Me	Me	A 1	СООН	CH ₂ CO	J4	0	СН
600	Ме	Ме	A 1	СООН	CH ₂ CO	J5	0	СН

Table 25

			•	able 25				
Compound No.	R ¹	R ²	Α	E	G	J	m	Х
601	Ме	Me	A1	соон	CH ₂ CO	J6	0	СН
602	Me	Ме	A1	соон	CH ₂ CO	J7	0	СН
603	Ме	Ме	A1	СООН	CH ₂ CO	J8	0	СН
604	Ме	Me	A1	СООН	CH ₂ CO	J9	0	СН
605	Ме	Me	A1	СООН	CH ₂ CO	J10	0	СН
606	Ме	Me	A1	СООН	CH ₂ CO	J11	0	СН
607	Me	Ме	A1	СООН	CH ₂ CO	J12	0	СН
608	Me	Ме	A1	СООН	CH2CO	J13	0	,CH
609	Me	Me	A1	соон	CH ₂ CO	J17	0	СН
610	Ме	Ме	A1	СООН	CH ₂ CO	J18	0	СН
611	Me	Me	A1	соон	CH ₂ CO	J19	0	СН
612	Me	Me	A1	СООН	CH ₂ CO	J23	0	СН
613	Ме	Me	A1	СООН	CH ₂ CO	J24	0	СН
614	Ме	Me	A1	СООН	CH ₂ CO	J25	0	СН
615	Me	Me	A1	соон	CH ₂ CO	J36	0	СН
616	Me	Me	A1	СООН	CH ₂ CO	J47	0	СН
617	Ме	Me	A1	СООН	CH ₂ CO	J57	0	СН
618	Me	Me	A1	СООН	CH ₂ CO	J62	0	СН
619	Н	Н	A1	СООН	CH ₂ CONH	J1	0	СН
620	Н	Ξ	A1	СООН	CH ₂ CONH	J2	0	СН

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Table 25 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	х
621	Н	Н	A1	соон	CH₂CONH	J3	0	СН
622	Н	Н	A1	соон	CH ₂ CONH	J4	0	СН
623	Н	Н	A1	соон	CH ₂ CONH	J5	0	СН
624	Н	Н	A1	СООН	CH ₂ CONH	J6	0	СН
625	Н	Н	A 1	СООН	CH ₂ CONH	J7	0	СН

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Table 26

			•	able 26				
Compound No.	R ¹	R ²	Α	E	G	J	m	Х
626	Н	Н	A 1	СООН	CH ₂ CONH	J8	0	СН
627	Н	н	A 1	СООН	CH₂CONH	J9	0	СН
628	Н	Н	A 1	COOH	CH₂CONH	J10	0	СН
629	Н	Н	A 1	COOH	CH₂CONH	J11	0	СН
630	Н	Н	A1	СООН	CH ₂ CONH	J12	0	СН
631	н	н	A1	СООН	CH ₂ CONH	J13	0	СН
632	Н	Н	A 1	СООН	CH ₂ CONH	J14	0	СН
633	Н	Н	A1	СООН	CH ₂ CONH	J15	0	СН
634	Н	Н	A 1	COOH	CH ₂ CONH	J16	0	СН
635	Н	н	A 1	СООН	CH ₂ CONH	J17	0	СН
636	н	Н	A 1	СООН	CH₂CONH	J18	0	СН
637	Н	Н	A1	СООН	CH ₂ CONH	J19	0	СН
638	Н	Н	A1	СООН	CH ₂ CONH	J20	0	CH
639	н	Н	A1	СООН	CH ₂ CONH	J21	0	СН
640	Н	Н	A1	COOH	CH ₂ CONH	J22	0	СН
641	Н	Н	A1	СООН	CH ₂ CONH	J23	0	СН
642	Н	Н	A 1	СООН	CH ₂ CONH	J24	0	СH
643	Н	Н	A 1	СООН	CH2CONH	J25°	0	СН
644	Н	Н	A1	соон	CH ₂ CONH	J26	0	СН
645	Н	Н	A1	СООН	CH₂CONH	J27	0	СН
646	Н	Н	A1	СООН	CH ₂ CONH	J28	0	СН
647	Н	Н	A1	СООН	CH₂CONH	J29	0	СН
648	Н	Н	A1	СООН	CH ₂ CONH	J30	0	СН
649	Н	Н	A1	соок	CH₂CONH	J31	0	СН
650	Н	Н	A1	СООН	CH₂CONH	J32	0	СН

Table 27

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			I	able 27				
Compound No.	R ¹	R ²	Α	E	G	J	m	Х
651	Н	Н	A1	СООН	CH ₂ CONH	J33	0	СН
652	Н	Н	A1	соон	CH ₂ CONH	J34	0	СН
653	Н	Н	A1	соон	CH ₂ CONH	J35	0	СН
654	Н	Н	A1	соон	CH ₂ CONH	J37	0	СН
655	Н	Н	A1	соон	CH ₂ CONH	J39	0	СН
656	Н	Н	A1	СООН	CH ₂ CONH	J62	0	СН
657	Н	Н	A1	СООН	CH ₂ CONH	J63	0	СН
658	Me	Me	A1	СООН	CH ₂ CONH	J1	0	СН
659	Ме	Me	A1	СООН	CH ₂ CONH	J2	0	СН
660	Ме	Me	A1	соон	CH ₂ CONH	J3	0	СН
661	Ме	Me	A1	СООН	CH ₂ CONH	J4	0	СН
662	Ме	Ме	A1	СООН	CH ₂ CONH	J5	0	СН
663	Ме	Ме	A1	СООН	CH ₂ CONH	J6	0	СН
664	Ме	Me	A1	СООН	CH ₂ CONH	J7	0	СН
665	Me	Me	A1	СООН	CH ₂ CONH	J8	0	СН
666	Ме	Me	A1	СООН	CH ₂ CONH	J9	0	СН
667	Me	Me	A 1	СООН	CH ₂ CONH	J10	0	СН
668	Me	Me	A1	СООН	CH ₂ CONH	J11	0	СН
669	Me	Me	A1	СООН	CH ₂ CONH	J12	0	СН
670	Me	Me	A1	СООН	CH ₂ CONH	J13	0	СН
671	Me	Me	A 1	СООН	CH ₂ CONH	J14	0	СН
672	Me	Me	A 1	СООН	CH ₂ CONH	J15	0	СН
673	Me	Me	A 1	соон	CH ₂ CONH	J16	0	СН
674	Me	Me	A 1	СООН	CH ₂ CONH	J17	0	СН
675	Me	Me	A1	СООН	CH ₂ CONH	J18	0	СН
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Compound No.	R ¹	R ²	Α	E	G	J	m	Х
676	Me	Me	A1	СООН	CH₂CONH	J19	0	СН
677	Ме	Me	A1	СООН	CH ₂ CONH	J20	0	СН
678	Me	Me	A1	СООН	CH₂CONH	J21	0	СН
679	Ме	Me	A1	СООН	CH ₂ CONH	J22	0	СН
680	Ме	Me	A1	соон	CH ₂ CONH	J23	0	СН
681	Ме	Me	A1	соон	CH ₂ CONH	J24	0	СН
682	Me	Me	A1	СООН	CH ₂ CONH	J25	0	СН

Table 28 (continued)

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Compound No.	R ¹	R ²	Α	E	G	J	m	Х
683	Ме	Me	A1	СООН	CH ₂ CONH	J26	0	СН
684	Ме	Me	A1	СООН	CH ₂ CONH	J27	0	СН
685	Me	Me	A1	СООН	CH ₂ CONH	J28	0	СН
686	Ме	Ме	A 1	СООН	CH ₂ CONH	J 29	0	СН
687	Me	Me	A1	СООН	CH ₂ CONH	J30	0	СН
688	Me	Ме	A1	СООН	CH ₂ CONH	J31	0	СН
689	Ме	Ме	A1	СООН	CH ₂ CONH	J32	0	СН
690	Ме	Ме	A1	СООН	CH ₂ CONH	J33	0	СН
691	Ме	Ме	A1	СООН	CH ₂ CONH	J34	0	СН
692	Ме	Me	A1	соон	CH ₂ CONH	J35	0	СН
693	Ме	Ме	A1	СООН	CH ₂ CONH	J37	0	СН
694	Ме	Ме	A1	СООН	CH ₂ CONH	J39	0	СН
695	Ме	Ме	A1	СООН	CH ₂ CONH	J62	0	СН
696	Ме	Me	A1	СООН	CH ₂ CONH	J63	0	СН
697	Н	Н	A1	СООН	CH ₂ CH ₂ O	J1	0	СН
698	Н	Н	A 1	СООН	CH ₂ CH ₂ O	J2	0	СН
699	Ι	Н	A 1	СООН	CH ₂ CH ₂ O	J3	0	СН
700	H	H	A 1	СООН	CH ₂ CH ₂ O	J4	0	СН

Table 29

Compound No.	R ¹	R ²	Α	E	G	J	Э	Х
701	Н	Н	A 1	соон	CH ₂ CH ₂ O	J5	0	СН
702	Н	Н	A 1	СООН	CH ₂ CH ₂ O	J6	0	СН
703	Н	Н	A1	СООН	CH ₂ CH ₂ O	J7	0	СН
704	Н	Н	A1	COCH	*CH2CH2O -	. 13	. 0_	СН
705	Н	н	A1	COOH	CH ₂ CH ₂ O	J9	0	СН
706	Н	Н	A 1	СООН	CH ₂ CH ₂ O	J10	0	СН
707	Н	Н	A1	соон	CH ₂ CH ₂ O	J11	0	СН
708	Н	Н	A1	соон	CH ₂ CH ₂ O	J12	0	СН
709	Н	Н	A1	соон	CH ₂ CH ₂ O	J13	0	СН
710	Н	Н	A1	соон	CH2CH2O	J14	0	СН
711	Н	Н	A1	СООН	CH ₂ CH ₂ O	J15	0	СН
712	Н	Н	A1	соон	CH ₂ CH ₂ O	J16	0	СН
713	Н	Н	A1	соон	CH ₂ CH ₂ O	J17	0	СН
714	Н	Н	A1	соон	CH ₂ CH ₂ O	J18	0	СН
715	Н	Н	A1	соон	CH ₂ CH ₂ O	J19	0	СН

Table 29 (continued)

Compound No.	R ¹	R ²	A	Е	G	J	m	х
716	I	H	A1	СООН	CH ₂ CH ₂ O	J20	0	СН
717	Н	Н	A1	СООН	CH ₂ CH ₂ O	J21	0	СН
718	Ι	H	A1	соон	CH ₂ CH ₂ O	J22	0	СН
719	Ι	H	A1	СООН	CH ₂ CH ₂ O	J23	0	СН
720	Ι	H	A1	СООН	CH ₂ CH ₂ O	J24	0	СН
721	Ħ	Н	A1	СООН	CH ₂ CH ₂ O	J25	0	СН
722	I	Н	A1	СООН	CH ₂ CH ₂ O	J26	0	СН
723	Н	H	A1	СООН	CH ₂ CH ₂ O	J27	0	СН
724	Ξ	Ξ	A1	СООН	CH ₂ CH ₂ O	J28	0	СН
725	Н	Н	A1	СООН	CH ₂ CH ₂ O	J29	0	CH

0		I 52		T =				
Compound No.	R ¹	R ²	Α	Е	G	J	m	X
726	Н	Н	A1	соон	CH ₂ CH ₂ O	J30	0	CH
727	Н	Н	A1	COOH	CH ₂ CH ₂ O	J31	0	СН
728	Н	Н	A1	соон	CH ₂ CH ₂ O	J32	0	СН
729	Н	Н	A1	СООН	CH ₂ CH ₂ O	J33	0	СН
730	H	Н	A1	соон	CH ₂ CH ₂ O	J34	0	СН
731	Н	Н	A1	соон	CH ₂ CH ₂ O	J35	0	СН
732	Н	Н	A1	соон	CH ₂ CH ₂ O	J37	0	СН
733	Н	Н	A1	СООН	CH ₂ CH ₂ O	J39	0	СН
734	Н	Н	A1	СООН	CH ₂ CH ₂ O	J62	0	СН
735	Н	Н	A1	СООН	CH ₂ CH ₂ O	J63	0	СН
736	Me	Me	A1	СООН	CH ₂ CH ₂ O	J1	0	СН
737	Ме	Me	A1	СООН	CH2CH2O	J2	0	СН
738	Me	Ме	AT	COOH	CH2CH2O	J3-	[®] O	СН
739	Me	Ме	A1	соон	CH2CH2O	J4	0	СН
740	Me	Ме	A1	СООН	CH ₂ CH ₂ O	J5	0	СН
741	Me	Ме	A1	соон	CH ₂ CH ₂ O	J6	0	СН
742	Ме	Ме	A1	СООН	CH ₂ CH ₂ O	J7	0	СН
743	Me	Ме	A1	СООН	CH ₂ CH ₂ O	J8	0	СН
744	Me	Ме	A1	СООН	CH ₂ CH ₂ O	J9	0	СН
745	Me	Me	A1	СООН	CH ₂ CH ₂ O	J10	0	СН
746	Ме	Me	A 1	соон	CH ₂ CH ₂ O	J11	0	СН
747	Me	Ме	A 1	СООН	CH ₂ CH ₂ O	J12	0	СН
748	Me	Me	A1	соон	CH ₂ CH ₂ O	J13	0	СН
749	Ме	Me	A 1	соон	CH ₂ CH ₂ O	J14	0	СН

Table 30 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	X
750	Me	Me	A1	СООН	CH ₂ CH ₂ O	J15	0	СН

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Table 31

			7	able 31				
Compound No.	R ¹	R ²	Α	E	G	J	m	х
751	Ме	Ме	A1	соон	CH ₂ CH ₂ O	J15	0	СН
752	Me	Ме	A1	соон	CH ₂ CH ₂ O	J16	0	СН
753	Me	Me	A1	соон	CH ₂ CH ₂ O	J17	0	СН
754	Ме	Me	A1	соон	CH ₂ CH ₂ O	J18	0	СН
755	Ме	Me	A1	СООН	CH ₂ CH ₂ O	J19	0	СН
756	Me	Me	A1	соон	CH ₂ CH ₂ O	J20	0	СН
757	Me	Me	A1	соон	CH ₂ CH ₂ O	J21	0	СН
758	Me	Ме	A1	СООН	CH ₂ CH ₂ O	J22	0	СН
759	Me	Me	A1	СООН	CH ₂ CH ₂ O	J23	0	СН
760	Me	Me	A1	СООН	CH ₂ CH ₂ O	J24	0	СН
761	Me	Me	A1	соон	CH ₂ CH ₂ O	J25	0	СН
762	Ме	Ме	A1	СООН	CH ₂ CH ₂ O	J26	0	СН
763	Me	Me	A1	СООН	CH ₂ CH ₂ O	J27	0	СН
764	Me	Me	A1	соон	CH ₂ CH ₂ O	J28	0	СН
765	Me	Ме	A1	СООН	CH ₂ CH ₂ O	J29	0	СН
766	Me	Me	A1	СООН	CH ₂ CH ₂ O	J30	0	СН
767	Ме	Ме	A1	COOH	CH ₂ CH ₂ O	J31	0	СН
768	Me	Ме	A 1	СООН	CH ₂ CH ₂ O	J32	0	СН
769	Me	Me	A 1	СООН	CH ₂ CH ₂ O	J33	0	СН
770	Me	Me	A 1	СООН	CH ₂ CH ₂ O	J34	0	СН
771	Me	Me	A 1	СООН	CH ₂ CH ₂ O	J35	0	CH
772	Me	Ме	ΑÏ	СООН	CH ₂ CH ₂ O	J37-	0	СН
773	Me	Ме	A1	соон	CH ₂ CH ₂ O	J39	0	СН
774	Me	Me	A1	соон	CH ₂ CH ₂ O	J62	0	СН
775	Ме	Me	A 1	соон	CH ₂ CH ₂ O	J63	0	СН

Compound No.	R ¹	R ²	Α	E	G	J	3	Х
776	Н	Ħ	A 1	СООН	CH ₂ S	J1	0	СН
777	Н	H	A 1	СООН	CH ₂ S	J2	0	СН
778	H	H	A 1	СООН	CH ₂ S	J3	0	СН

Table 32 (continued)

Compound No.	R ¹	R ²	A	E	G	J	m	х
			<u> </u>					
779	Н	Н	A1	соон	CH ₂ S	J4	0	СН
780	Н	Н	A1	СООН	CH ₂ S	J8	0	СН
781	Н	Н	A1	СООН	CH ₂ S	J9	0	СН
782	Н	Н	A1	соон	CH ₂ S	J10	0	СН
783	Ме	Ме	A1	СООН	CH ₂ S	J1	0	СН
784	Ме	Me	A1	СООН	CH ₂ S	J2	0	СН
785	Me	Me	A1	СООН	CH ₂ S	J3	0	СН
786	Ме	Me	A1	СООН	CH ₂ S	J4	0	СН
787	Me	Me	A1	СООН	CH ₂ S	J8	0	СН
788	Ме	Ме	A1	соон	CH ₂ S	J9	0	СН
789	Ме	Me	A1	СООН	CH ₂ S	J10	0	СН
790	Н	Н	A1	СООН	CH ₂ SO ₂	J1	0	СН
791	Н	Н	A1	СООН	CH ₂ SO ₂	J2	0	СН
792	Н	Н	A1	СООН	CH ₂ SO ₂	JЗ	0	СН
793	Н	Η	A1	СООН	CH ₂ SO ₂	J4	0	СН
794	Н	Н	A1	СООН	CH ₂ SO ₂	J8	0	СН
795	Н	Н	A 1	СООН	CH ₂ SO ₂	J9	0	СН
796	Н	Н	A 1	СООН	CH ₂ SO ₂	J10	0	СН
797	Me	Me	A 1	СООН	CH ₂ SO ₂	J1	0	СН
798	Ме	Me	A1	СООН	CH ₂ SO ₂	J2	0	СН
799	Ме	Me	A 1	COOH	CH ₂ SO ₂	J3	0	СН
800	Ме	Ме	A 1	СООН	CH ₂ SO ₂	J4	0	СН

Compound No.	-91.: -81.:	≥ R2=	Nº A	TEST.	III.	J	m	Х
801	Ме	Ме	A 1	СООН	CH ₂ SO ₂	J8	0	СН
802	Me	Me	A 1	СООН	CH ₂ SO ₂	J9	0	СН
803	Me	Ме	A1	СООН	CH ₂ SO ₂	J10	0	СН
804	Ме	Ме	A1	СООН	CH ₂	J81	0	СН
805	Me	Ме	A1	СООН	CH ₂	J82	٥	СН
806	Me	Me	A1	СООН	CH ₂	J83	0	СН
807	Ме	Ме	A 1	СООН	CH ₂	J84	0	СН
808	Ме	Me	A 1	СООН	CH ₂	J85	0	СН
809	Н	Н	A1	СООН	CH ₂	J81	0	СН
810	H	Н	A1	СООН	CH ₂	J82	0	СН
811	Н	Н	A1	СООН	CH ₂	J83	0	СН

Table 33 (continued)

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Compound No.	R ¹	R ²	A	E	G	J	m	Х
812	Н	Н	A1	СООН	CH ₂	J84	0	СН
813	Н	Н	A1	СООН	CH ₂	J85	0	СН
814	Ме	Me	A1	СООН	CH ₂ CH ₂	J1	1	СН
815	Me	Me	A1	СООН	CH ₂	J1	1	СН
816	Ме	Me	A1	СООН	CH ₂	J37	1	СН
817	Ме	Me	A1	СООН	CH ₂	J39	1	СН
818	Ме	Me	A1	СООН	CH ₂	J50	1	СН
819	Ме	Me	A1	СООН	CH ₂	J63	1	СН
820	Ме	Me	A1	СООН	CH ₂	J64	1	СН
821	Ме	Ме	A1	СООН	CH ₂	J65	1	СН
822	Н	Н	A1	СООН	CH ₂	J37	1	СН
823	Н	H	A1	СООН	CH ₂	J39	1	СН
824	Н	Н	A1	СООН	CH ₂	J50	1	СН
825	Н	Н	A1	СООН	CH ₂	J63	1	СН

Table 34

Compound No.	R ¹	R ²	Α	Е	G	J	m	Х
826	Н	Н	A1	СООН	CH ₂	J64	1	СН
827	Н	Н	A 1	СООН	CH ₂	J65	1	СН
828	CI	ច	A 1	СООН	CH ₂	J37	1	СН
829	CI	CI	A 1	соон	CH ₂	J39	1	СН
830	CI	CI	A1	СООН	CH ₂	J50	1	СН
831	CI	CI	A1	СООН	CH ₂	J63	1	СН
832	CI	CI	A 1	СООН	CH ₂	J64	1	СН
833	CI	CI	A 1	COOH	CH ₂	J65	1	СН
834	H	H	A4	СООН	ECH ₂	J37	1	СН
835	Н	Н	A 4	соон	CH ₂	J39	1	СН
836	Н	Н	A 4	СООН	CH ₂	J50	1	СН
837	Н	Н	A4	соон	CH ₂	J63	1	СН
838	Н	Н	A4	соон	CH ₂	J64	1	СН
839	Н	Н	A4	соон	CH ₂	J65	1	СН
840	Н	Н	A11	соон	CH ₂	J37	1	СН
841	Н	Н	A11	соон	CH ₂	J39	1	СН
842	Н	Н	A11	соон	CH ₂	J50	1	СН
843	Н	Н	A11	соон	CH ₂	J63	1	СН
844	Н	Н	A11	соон	CH ₂	J64	1	СН
845	Н	Н	A11	СООН	CH ₂	J65	1	СН

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Table 34 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
846	Η	Н	A18	соон	CH ₂	J37	1	СН
847	Н	Н	A18	СООН	CH ₂	J39	1	СН
848	Н	Н	A18	СООН	CH ₂	J50	1	СН
849	H	Н	A18	СООН	CH ₂	J63	1	СН
850	Н	Н	A18	соон	CH ₂	J64	1	СН

			Tal	ble 35				
Compound No.	R ¹	R ²	Α	E	G	J	m	X
851	Н	Н	A18	соон	CH ₂	J65	1	СН
852	Н	Н	A20	соон	CH ₂	J37	1	СН
853	Н	Н	A20	соон	CH ₂	J39	1	СН
854	Н	Н	A20	соон	CH ₂	J50	1	СН
855	Н	Н	A20	соон	CH ₂	J63	1	СН
856	Н	Н	A20	соон	CH ₂	J64	1	СН
857	Н	Н	A20	соон	CH ₂	J65	1	СН
858	Ме	Ме	A 1	соон	CH ₂ CH ₂	J1	2	СН
859	Ме	Me	A1	соон	CH ₂	J1	2	СН
860	Ме	Ме	A1	соон	CH ₂	J37	2	СН
861	Ме	Ме	A1	соон	CH ₂	J 39	2	СН
862	Ме	Ме	A1	соон	CH ₂	J50	2	СН
863	Ме	Ме	A1	соон	CH ₂	J63	2	СН
864	Ме	Ме	A1	соон	CH ₂	J64	2	СН
865	Me	Me	A1	соон	CH ₂	J65	2	СН
866	Н	Н	A1	соон	CH ₂	J37	2	СН
867	<u>-H-</u>	Н.	A1	COOH	_{}≿} CH ₂	J39	2	СН
868	Н	H	A1	СООН	CH ₂	J50	2	СН
869	Н	Н	A1	СООН	CH ₂	J63	2	СН
870	Н	Н	A1	СООН	CH ₂	J64	2	СН
871	Н	Н	A1	СООН	CH ₂	J65	2	СН
872	ਹ	а	A 1	СООН	CH ₂	J37	2	СН
873	ō	CI	A1	СООН	CH ₂	J39	2	СН
874	CI	CI	A1	соон	CH ₂	J50	2	СН
875	CI	CI	A1	СООН	СНэ	J63	2	СН

Table 36

10	

			IBDI	E 30				
Compound No.	R ¹	R ²	A	Е	G	J	m	Х
876	CI	CI	A1	соон	CH ₂	J64	2	СН
877	CI	CI	A1	соон	CH ₂	J65	2	СН
878	Н	Н	A1	соон	CH ₂	J37	2	N
879	Н	Н	A1	соон	CH ₂	J39	2	N
880	Н	Н	A1	соон	CH ₂	J50	2	N
881	Н	Н	A1	СООН	CH ₂	J63	2	N
882	Н	Н	A1	COOH	CH ₂	J64	2	N
883	Н	Н	A1	соон	CH ₂	J65	2	N
884	Me	Н	A1	соон	CH ₂	J37	2	СН
885	Me	Н	A1	соон	CH ₂	J63	2	СН
886	Ме	Н	A1	соон	CH ₂	J64	2	СН
887	Ме	Н	A 1	COOH	CH ₂	J65	2	СН
888	Н	Н	A4	СООН	CH ₂	J37	2	СН
889	Н	Н	A4	СООН	CH ₂	J63	2	СН
890	Н	Н	A4	соон	CH ₂	J64	2	СН
891	Н	Н	A 4	СООН	CH ₂	J65	2	СН
892	Ме	Me	A4	соон	CH ₂	J37	2	СН
893	Ме	Me	A4	соон	CH ₂	J63	2	СН
894	Ме	Ме	A4	COOH	CH ₂	J64	2	СН
895	Ме	Ме	A4	соон	CH ₂	J65	2	СН
896	CI	CI	A4	соон	CH ₂	J37	2	СН
897	cī	CI	A4	соон	CH ₂	J63	2	СН
898	CI	а	A4	соон	CH ₂	J64	2	СН
899	CI	ប	A 4	соон	CH ₂	J65	2	СН
900	Н	Н	A4	соон	CH ₂	J37	2	N
The same of the same of the		زنتينج	7					

Table 37

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
901	Н	Н	A4	СООН	CH ₂	J63	2	N
902	Н	Н	A4	СООН	CH ₂	J64	2	N
903	H	Н	A4	СООН	CH ₂	J65	2	N
904	H	Н	A11	СООН	CH ₂	J37	2	СН
905	H	Н	A11	СООН	CH ₂	J63	2	СН
906	H	Н	A11	СООН	CH ₂	J64	2	СН
907	Н	Н	A11	СООН	CH ₂	J65	2	СН

Table 37 (continued)

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Compound No.	R ¹	R ²	Α	Ε	G	J	m	Х
908	Me	Ме	A11	соон	CH ₂	J37	2	СН
909	Ме	Ме	A11	соон	CH ₂	J63	2	СН
910	Ме	Ме	A11	соон	CH ₂	J64	2	СН
911	Ме	Me	A11	соон	CH ₂	J65	2	СН
912	CI	CI	A11	соон	CH ₂	J37	2	СН
913	CI	ច	A11	соон	CH ₂	J63	2	СН
914	CI	C	A11	СООН	CH ₂	J64	2	СН
915	CI	C	A11	СООН	CH ₂	J65	2	СН
916	I	H	A11	соон	CH ₂	J37	2	N
917	H	Н	A11	СООН	CH ₂	J63	2	N
918	Ξ	Н	A11	соон	CH ₂	J64	2	N
919	H	н	A11	СООН	CH ₂	J65	2	N
920	Me	Me	A18	СООН	CH ₂	J37	2	СН
921	Ме	Me	A18	СООН	CH ₂	J63	2	СН
922	Me	Me	A18	СООН	CH ₂	J64	2	СН
923	Me	Ме	A18	соон	CH ₂	J65	2	CH
924	Н	Н	A18	СООН	CH ₂	J37	2	СН
925	Н	Н	A18	СООН	CH ₂	J63	2	СН

Table 38

Compound No.	R ¹	R ²	Α	E	G	j	m	Х
926	Н	Н	A18	соон	CH ₂	J64	2	СН
927	Н	Н	A18	соон	CH ₂	J65	2	СН
928	CI	CI	A18	соон	CH ₂	J37	2	СН
929	.ÇI.	–Cl ^z	A18	соон	CH ₂	J63	2	СН
930	CI	C	A18	соон	CH ₂	J64	2	CH
931	C	CI	A18	СООН	CH ₂	J65	2	СН
932	Н	Н	A18	соон	CH ₂	J37	2	N
933	Н	Н	A18	соон	CH ₂	J63	2	N
934	Н	Н	A18	соон	CH ₂	J64	2	N
935	Н	Н	A18	соон	CH ₂	J65	2	N
936	Me	Me	A20	соон	CH ₂	J37	2	СН
937	Ме	Ме	A 20	соон	CH ₂	J63	2	СН
938	Me	Me	A20	соон	CH ₂	J64	2	СН
939	Me	Me	A20	СООН	CH ₂	J65	2	СН
940	Н	Н	A20	СООН	CH ₂	J37	2	СН

Table 38 (continued)

Compound No.	R ¹	R ²	Α	Ε	G	J	m	Х
941	Н	Н	A20	СООН	CH ₂	J63	2	СН
942	Н	Н	A20	СООН	CH ₂	J64	2	СН
943	Н	Н	A20	СООН	CH ₂	J65	2	СН
944	CI	CI	A20	СООН	CH ₂	J37	2	СН
945	CI	CI	A20	соон	CH ₂	J63	2	СН
946	C	CI	A20	СООН	CH ₂	J64	2	СН
947	ច	C	A20	СООН	CH ₂	J65	2	СН
948	Ι	Ι	A20	СООН	CH ₂	J37	2	N
949	Н	Н	A20	СООН	CH ₂	J63	2	N
950	н	Н	A20	СООН	CH ₂	J64	2	N

Table 39

Compound No.	R ¹	R ²	Α	E	G	J	æ	Х
951	Н	Н	A20	СООН	CH ₂	J65	2	N
952	Me	Me	A 1	tetrazol	CH ₂	J37	0	СН
953	Me	Me	A1	tetrazol	CH ₂	J63	0	СН
954	Ме	Me	A1	tetrazol	CH ₂	J64	0	СН
955	Ме	Me	A1	tetrazol	CH ₂	J65	0	СН
956	Н	Н	A 1	tetrazol	CH ₂	J37	0	СН
957	Н	Н	A1	tetrazol	CH ₂	J63	0	СН
958	Н	Н	A 1	tetrazol	CH ₂	J64	0	СН
959	Н	Н	A1	tetrazol	CH ₂	J65	0	СН
960	CI	CI	A1	tetrazol	CH ₂	J37	0	СН
961	C	Ci	A1	tetrazol	CH ₂	J63	0	СН
962	ÇI	CI	A 1	tetrazol	CH ₂	J64	0	СН
963	CI	ČI	A1	tetrazol	CH ₂	J65	0	СН
964	Н	Н	A 1	tetrazol	CH ₂	J37	0	N
965	Н	Н	A 1	tetrazol	CH ₂	J63	0	N
966	Н	Н	A1	tetrazol	CH ₂	J64	0	N
967	Н	Н	A1	tetrazol	CH ₂	J65	0	N
968	Н	Н	A4	tetrazol	CH ₂	J37	0	СН
969	Н	Н	A4	tetrazol	CH ₂	J63	0	СН
970	Н	Н	A4	tetrazol	CH ₂	J64	0	СН
971	Н	Н	A4	tetrazol	CH ₂	J65	0	СН
972	Н	Н	A18	tetrazol	CH ₂	J37	0	СН
973	Н	Н	A18	tetrazol	CH ₂	J63	0	СН
974	Н	Н	A18	tetrazol	CH ₂	J64	0	СН

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Table 39 (continued)

Compound No.	R ¹	R ²	Α	E	G	J	m	Х
975	Η	Ι	A18	tetrazoi	CH ₂	J65	0	СН

Table 40

Compound No. R¹ R² A E G J m X 976 Me Me A19 tetrazol CH2 J37 0 CH 977 Me Me A19 tetrazol CH2 J63 0 CH 978 Me Me A19 tetrazol CH2 J64 0 CH 979 Me Me A19 tetrazol CH2 J65 0 CH 980 H H A19 tetrazol CH2 J63 0 CH 981 H H A19 tetrazol CH2 J63 0 CH 982 H H A19 tetrazol CH2 J65 0 CH 983 H H A19 tetrazol CH2 J63 0 CH 985 Me Me A20 tetrazol CH2 J64 0				labi	e 40				
977 Me Me A19 tetrazol CH ₂ J63 0 CH 978 Me Me A19 tetrazol CH ₂ J64 0 CH 979 Me Me A19 tetrazol CH ₂ J65 0 CH 980 H H A19 tetrazol CH ₂ J37 0 CH 981 H H A19 tetrazol CH ₂ J63 0 CH 982 H H A19 tetrazol CH ₂ J64 0 CH 983 H H A19 tetrazol CH ₂ J65 0 CH 984 Me Me A20 tetrazol CH ₂ J37 0 CH 985 Me Me A20 tetrazol CH ₂ J63 0 CH 986 Me Me A20 tetrazol CH ₂ J63 0 CH 987 Me Me A20 tetrazol CH ₂ J63 0 CH 988 H H A20 tetrazol CH ₂ J65 0 CH 989 H H A20 tetrazol CH ₂ J65 0 CH 989 H H A20 tetrazol CH ₂ J63 0 CH	Compound No.	R ¹	R ²	Α	E	G	J	m	Х
978	976	Ме	Ме	A19	tetrazol	CH ₂	J37	0	СН
979 Me Me A19 tetrazol CH ₂ J65 0 CH 980 H H A19 tetrazol CH ₂ J37 0 CH 981 H H A19 tetrazol CH ₂ J63 0 CH 982 H H A19 tetrazol CH ₂ J64 0 CH 983 H H A19 tetrazol CH ₂ J65 0 CH 984 Me Me A20 tetrazol CH ₂ J37 0 CH 985 Me Me A20 tetrazol CH ₂ J63 0 CH 986 Me Me A20 tetrazol CH ₂ J63 0 CH 987 Me Me A20 tetrazol CH ₂ J63 0 CH 988 H H A20 tetrazol CH ₂ J65 0 CH 989 H H A20 tetrazol CH ₂ J65 0 CH 990 H H A20 tetrazol CH ₂ J65 0 CH	977	Me	Ме	A19	tetrazol	CH ₂	J63	0	СН
980 H H A19 tetrazol CH ₂ J37 0 CH 981 H H A19 tetrazol CH ₂ J63 0 CH 982 H H A19 tetrazol CH ₂ J64 0 CH 983 H H A19 tetrazol CH ₂ J65 0 CH 984 Me Me A20 tetrazol CH ₂ J37 0 CH 985 Me Me A20 tetrazol CH ₂ J63 0 CH 986 Me Me A20 tetrazol CH ₂ J63 0 CH 987 Me Me A20 tetrazol CH ₂ J64 0 CH 988 H H A20 tetrazol CH ₂ J65 0 CH 989 H H A20 tetrazol CH ₂ J65 0 CH 990 H H A20 tetrazol CH ₂ J63 0 CH	978	Me	Me	A19	tetrazol	CH ₂	J64	0	СН
981 H H A19 tetrazol CH ₂ J63 0 CH 982 H H A19 tetrazol CH ₂ J64 0 CH 983 H H A19 tetrazol CH ₂ J65 0 CH 984 Me Me A20 tetrazol CH ₂ J37 0 CH 985 Me Me A20 tetrazol CH ₂ J63 0 CH 986 Me Me A20 tetrazol CH ₂ J63 0 CH 987 Me Me A20 tetrazol CH ₂ J64 0 CH 988 H H A20 tetrazol CH ₂ J65 0 CH 989 H H A20 tetrazol CH ₂ J63 0 CH 990 H H A20 tetrazol CH ₂ J63 0 CH	979	Me	Me	A19	tetrazol	CH ₂	J65	0	СН
982 H H A19 tetrazol CH ₂ J64 0 CH 983 H H A19 tetrazol CH ₂ J65 0 CH 984 Me Me A20 tetrazol CH ₂ J37 0 CH 985 Me Me A20 tetrazol CH ₂ J63 0 CH 986 Me Me A20 tetrazol CH ₂ J64 0 CH 987 Me Me A20 tetrazol CH ₂ J65 0 CH 988 H H A20 tetrazol CH ₂ J37 0 CH 989 H H A20 tetrazol CH ₂ J37 0 CH 990 H H A20 tetrazol CH ₂ J63 0 CH	980	Н	Н	A19	tetrazol	CH ₂	J37	0	СН
983 H H A19 tetrazol CH2 J65 0 CH 984 Me Me A20 tetrazol CH2 J37 0 CH 985 Me Me A20 tetrazol CH2 J63 0 CH 986 Me Me A20 tetrazol CH2 J64 0 CH 987 Me Me A20 tetrazol CH2 J65 0 CH 988 H H A20 tetrazol CH2 J37 0 CH 989 H H A20 tetrazol CH2 J63 0 CH 990 H H A20 tetrazol CH2 J64 0 CH	981	Н	Н	A19	tetrazol	CH ₂	J63	0	СН
984 Me Me A20 tetrazol CH ₂ J37 0 CH 985 Me Me A20 tetrazol CH ₂ J63 0 CH 986 Me Me A20 tetrazol CH ₂ J64 0 CH 987 Me Me A20 tetrazol CH ₂ J65 0 CH 988 H H A20 tetrazol CH ₂ J37 0 CH 989 H H A20 tetrazol CH ₂ J63 0 CH 990 H H A20 tetrazol CH ₂ J63 0 CH	982	Н	Н	A19	tetrazol	CH ₂	J64	0	СН
985 Me Me A20 tetrazol CH2 J63 0 CH 986 Me Me A20 tetrazol CH2 J64 0 CH 987 Me Me A20 tetrazol CH2 J65 0 CH 988 H H A20 tetrazol CH2 J37 0 CH 989 H H A20 tetrazol CH2 J63 0 CH 990 H H A20 tetrazol CH2 J64 0 CH	983	Н	Н	A19	tetrazol	CH ₂	J65	0	СН
986 Me Me A20 tetrazol CH ₂ J64 0 CH 987 Me Me A20 tetrazol CH ₂ J65 0 CH 988 H H A20 tetrazol CH ₂ J37 0 CH 989 H H A20 tetrazol CH ₂ J63 0 CH 990 H H A20 tetrazol CH ₂ J64 0 CH	984	Me	Me	A20	tetrazol	CH ₂	J37	0	СН
987 Me Me A20 tetrazol CH2 J65 0 CH 988 H H A20 tetrazol CH2 J37 0 CH 989 H H A20 tetrazol CH2 J63 0 CH 990 H H A20 tetrazol CH2 J64 0 CH	985	Me	Me	A20	tetrazol	CH ₂	J63	0	СН
988 H H A20 tetrazol CH2 J37 0 CH 989 H H A20 tetrazol CH2 J63 0 CH 990 H H A20 tetrazol CH2 J64 0 CH	986	Me	Me	A20	tetrazol	CH ₂	J64	0	СН
989 H H A20 tetrazol CH ₂ J63 0 CH 990 H H A20 tetrazol CH ₂ J64 0 CH	987	Me	Me	A20	tetrazol	CH ₂	J65	0	СН
990 H H A20 tetrazol CH ₂ J64 0 CH	988	Н	Н	A20	tetrazol	CH ₂	J37	0	CH
	989	Н	Н	A20	tetrazol	CH ₂	J63	0	СН
991 H H A20 tetrazol CH ₂ J65 0 CH	990	Н	Н	A20	tetrazol	CH ₂	J64	0	СН
	991	Н	Н	A20	tetrazol	CH ₂	J65	0	СН

[0021] The thiobenzimidazole derivative (1) of the present invention in which E is COOH and m is 0 can be prepared by the synthetic method (A) or (B) shown below:

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Synthetic method (A)

[0022]

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$$\begin{array}{c|c}
Z = G \\
\hline
 & R^{1}
\end{array}$$

$$\begin{array}{c|c}
R^{1}
\end{array}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}$$

$$\begin{array}{c|c}
R^{2}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}$$

$$\begin{array}{c|c}
R^{2}
\end{array}$$

$$\begin{array}{c|c}
R^{2}$$

wherein Z represents a halogen, R1, R2, R3, A, G, J, and X are as defined above.

[0023] Thus, the nitro group of a 2-nitroaniline derivative (a1) is reduced to give an orthophenylene diamine (a2). CS₂ is reacted with this diamine to produce a compound (a3), with which a halide ester derivative (a4) is reacted to obtain (a5). A halide derivative (a6) is reacted therewith to obtain (a7), which is hydrolyzed to yield a benzimidazole derivative (a8) of the present invention.

[0024] The reduction of the nitro group may be carried out under a standard condition for catalytic reduction. For example, a reaction is carried out with hydrogen gas in the presence of a catalyst such as Pd-C at a temperature of room temperature to 100°C. Alternatively, a method of treatment using zinc or tin under an acidic condition, or a method of using zinc powder at a neutral or alkaline condition can be used.

[0025] The reaction of an orthophenylene diamine derivative (a2) with CS₂ may be carried out using, for example, a method as described in J. Org. Chem. 19: 631-637, 1954, or J. Med. Chem. 36: 1175-1187, 1993 (EtOH solution).

[0026] The reaction of a thiobenzimidazole (a3) and a halide ester (a4) may be carried out according to the condition of the conventional S-alkylation, for example in the presence of a base such as NaH, Et₃N, NaOH, or K₂CO₃ at a temperature of 0°C to 200°C under stirring.

[0027] The reaction of a thiobenzimidazole (a5) and a halide derivative (a6) may be carried out according to the condition for the conventional N-alkylation or N-acylation, for example in the presence of a base such as NaH, Et₃N, NaOH, or K₂CO₃ at a temperature of 0°C to 200°C under stirring.

5 [0028] As the elimination reaction of the carboxy protecting group R³, preferably a method of hydrolysis is employed using an alkali such as lithium hydroxide or an acid such as trifluoroacetic acid.

Synthetic method (B)

[0029]

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[0030] Thus, the amino group of a 2-nitroaniline derivative (a1) can be protected with L to give (b1). A halide derivative (a6) is reacted therewith to obtain (b2), from which L is deprotected to obtain (b3). The nitro group of (b3) is reduced to obtain an orthophenylene diamine derivative (b4). CS₂ is reacted therewith to yield a compound (b5), with which a halide ester derivative (a4) is reacted to obtain (a7) which may be hydrolyzed to yield a benzimidazole derivative of the present invention. Alternatively, it is also possible to obtain a compound (b3) directly by allowing the 2-nitroaniline derivative (a1) as it is unprotected to be reacted to a halide derivative (a6) or an aldehyde derivative (b6). As the protecting group L, there can be mentioned a trifluoroacetic acetyl group, an acetyl group, a t-butoxycarbonyl group, a benzyl group, and the like. The reaction of the 2-nitroaniline derivative (a1) and the aldehyde derivative (b6) may be carried out according to the conditions of the conventional reductive amination using a reducing agent such as a complex hydrogen compound, for example LiAlH₄, NaBH₄, NaBH₆OAc)₃, etc. or diborane, in a solvent such as ethanol, methanol, and dichloromethane at a temperature condition of 0°C to 200°C. The other reactions may be carried out as in the Synthetic method (A).

[0031] The thiobenzimidazole derivative (1) of the present invention in which E is COOH, m is 0, and G is an amide bond can be prepared by the synthetic method (C) shown below:

Synthetic method (C)

[0032]

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5 z^Q~cootBu COOR³ COOR (c1) 10 **COOtBu** (a5) (c2) COOR COOR3 J-NH. (c4)15 HOOC 20 (c3)

wherein Q represents a methylene group, a phenylene group, etc., and Z represents a halogen. R¹, R², R³, A, J, and X are as defined above, provided that R³ is a protecting group such as an ethyl group, a methyl group, etc. inactive in an acid.

[0033] Thus, a tert-butyl ester halide derivative (c1) is reacted with a thiobenzimidazole compound (a5) to obtain a compound (c2), which is subjected to hydrolysis under an acidic condition to yield (c3). An amine derivative (c4) is reacted therewith to yield (c5), which is subjected to hydrolysis to obtain the benzimidazole derivative of the present invention.

[0034] The condensation amidation may be carried out by a conventional method using a condensing agent. As the condensing agent, there can be mentioned DCC, DIPC, EDC=WSCI, WSCI • HCI, BOP, DPPA, etc., which may be used alone or in combination with HONSu, HOBt, HOOBt, etc. The reaction may be carried out in a appropriate solvent such as THF, chloroform, t-butanol, etc. at a temperature condition of 0°C to 200°C. The other reactions may be carried out as in the Synthetic method (A).

[0035] The thiobenzimidazole derivative (1) of the present invention in which E is COOH, m is 0, and G is an ether bond can be prepared by the synthetic method (D) shown below:

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Synthetic method (D)

[0036]

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wherein Z represents a halogen, R1, R2, R3, A, J, and X are as defined above.

[0037] Thus, a thiobenzimidazole compound (a5) is reacted with, for example, a halide alcohol derivative (d1) to yield a compound (d2). A phenol derivative (d3) is reacted therewith to yield an ether (d4), which is subjected to hydrolysis to yield a benzimidazole derivative (a8) of the present invention.

[0038] The etherification may be carried out using a phosphine compound such as triphenyl phosphine and tributyl phosphine and an azo compound such as DEAD and TMAD in a suitable solvent such as N-methylmorpholine and THF at a temperature of 0°C to 200°C in a Mitsunobu reaction or a related reaction thereof. The other reactions may be carried out as in the Synthetic method (A).

[0039] The thiobenzimidazole derivative (1) of the present invention in which E is a tetrazole and m is 0 can be prepared by the synthetic method (E) shown below:

Synthetic method (E)

[0040]

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R1

$$R$$
1

 R 2

 R 1

 R 2

 R 3

 R 4

 R 3

 R 4

 R 5

 R 5

 R 7

 R 7

 R 8

 R 9

 R

wherein R¹, R², A, G, J, and X are as defined above.

[0041] A nitrile (e1) is reacted with various azi compounds to be converted to a tetrazole (e2).

[0042] As the azi compound, there can be mentioned a trialkyltin azide compound such as trimethyltin azide, and hydrazoic acid or an ammonium salt thereof. When an organic tin azide compound is used, 1-4 fold molar amount is used relative to the compound (e1). When hydrazoic acid or an ammonium salt thereof is used, 1-5 fold molar amount of sodium azide or a tertiary amine such as ammonium chloride and triethylamine may be used relative to the com-

pound (e1). Each reaction may be carried out at at temperature of 0°C to 200°C in a solvent such as toluene, benzene and DMF.

[0043] The thiobenzimidazole derivative (1) of the present invention in which m is 1 or 2 can be prepared by the synthetic method (F) shown below:

Synthetic method (F)

[0044]

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R² X N S COOR³ (a7)

R¹ X N S O

RI COORS

wherein R1, R2, R3, A, G, J, and X are as defined above.

(f1)

[0045] Thus, a thiobenzimidazole compound (a7) may be reacted with a peroxide compound in a suitable medium to yield a sulfoxide derivative (f1) and/or a sulfone derivative (f2). As the peroxide compound used, there can be mentioned perbenzoic acid, m-chloroperbenzoic acid, peracetic acid, hydrogeny peroxide, and the like, and as the solvent used, there can be mentioned chloroform, dichloromethane, and the like. The ratio of the compound (a7) to the peroxide compound used is selected from, but not limited to, a broad range as appropriate, and generally 1.2 to 5 fold molar amount, for example, may be preferably used. Each reaction is carried out generally at about 0 to 50°C, and preferably at 0°C to room temperature, and is generally complete in about 4-20 hours.

[0046] The benzimidazole derivatives of the present invention can be converted, as needed, to medically acceptable non-toxic cation salts. As such a salt, there can be mentioned an alkali metal ion such as Na⁺ and K⁺; an alkaline earth metal ion such as Mg²⁺ and Ca²⁺; a metal ion such as Al³⁺ and Zr²⁺; or an organic base such as ammonia, triethylamine, ethylenediamine, propanediamine, pyrrolidine, piperidine, piperadine, pyridine, lysine, choline, eth-canciamine, N,N-diethylethanolamine, 4-hydroxypiperidine, glucosamine, and N-methylglucamine. Among them, Na⁺, Ca²⁺, lysine, choline, N,N-dimethylethanolamine and N-methylglucamine are preferred.

[0047] The benzimidazole derivatives of the present invention inhibit human chymase activity. Specifically, their IC50 is not greater than 1000, preferably not smaller than 0.01 and less than 1000, and more preferably not smaller than 0.05 and less than 500. The benzimidazole derivatives of the present invention having such excellent inhibitory action on human chymase can be used as clinically applicable preventive and/or therapeutic agents for various diseases.

[0048] The benzimidazole derivatives of the present invention can be administered as pharmaceutical compositions together with pharmaceutically acceptable carriers by oral or parenteral routes after being shaped into various dosage forms. As the parenteral administration, there can be mentioned intravenous, subcutaneous, intramuscular, percutaneous, rectal, nasal, and eye drop administration.

[0049] Dosage forms for said pharmaceutical compositions include the following. For example, in the case of oral administration, there can be mentioned dosage forms such as tablets, pills, granules, powders, solutions, suspensions, syrups, and capsules.

[0050] As used herein, tablets are shaped by a conventional method using a pharmaceutically acceptable carrier such as an excipient, a binder, and a disintegrant. Pills, granules, and powders can also be shaped by a conventional method using an excipient etc. Solutions, suspensions, and syrups may be shaped by a conventional method using glycerin esters, alcohols, water, vegetable oils, and the like. Capsules can be shaped by filling a granule, a powder, and a solution into a capsule made of gelatin etc.

[0051] Among the parenteral preparations, those for intravenous, subcutaneous, and intramuscular administration can be administered as an injection. As injections, a benzoic acid derivative is dissolved in a water soluble liquid such as physiological saline, or in a non-water soluble liquid comprising an organic ester such as propylene glycol, polyethylene glycol, and a vegetable oil.

In the case of percutaneous administration, dosage forms such as ointments and creams can be used. Ointments can be prepared by mixing a benzoic acid derivative with a fat or lipid, vaseline, etc., and creams can be prepared by mixing a benzoic acid derivative with an emulsifier.

[0053] In the case of rectal administration, gelatin soft capsules can be used to prepare suppositories.

[0054] In the case of nasal administration, they can be used as a formulation comprising a liquid or powder composition. As the base for liquid formulations, water, saline, a phosphate buffer, an acetate buffer etc. can be used, and furthermore they may include a surfactant, an antioxidant, a stabilizer, a preservative, and a thickening agent. As the base for powder formulations, there can be mentioned polyacrylic acid salts that are readily solubule in water, cellulose lower alkyl ethers, polyethylene glycol, polyvinylpyrrolidone, amylose, pullulan, etc. that are water-absorptive, or celluloses, starches, proteins, gums, crosslinked vinyl polymers, etc. that are hardly water-soluble, and preferably they are water-absorptive. Alternatively, they may be combined. Furthermore, for powder formulations, an antioxidant, a colorant, a preservative, a disinfectant, a corrigent, etc. can be added. Such liquid formulations arid powder formulations can be administered using, for example, a spraying device etc.

[0055] For eye drop administration, they can be used as aqueous or non-aqueous eye drops. For the aqueous eye drops, sterile purified water, physiological saline etc. can be used as a solvent. When sterile purified water is used as the solvent, a suspending agent such as a surf actant and a polymer thickener may be added to prepare an aqueous eye drop suspension. Alternatively, a solubilizing agent such as a nonionic surf actant may be added to prepare a soluble eye drop solution. The non-aqueous eye drop can use a non-aqueous solvent for injection as a solvent, and can be used as a non-aqueous eye drop solution.

[0056] In the case where administration to the eye is performed by a method other than the eye drop, dosage forms such as an eye ointment, an application solution, an epipastic, and an insert can be used.

[0057] In the case of nasal or oral inhalation, they are inhaled as a solution or a suspension of the benzimidazole derivatives of the present invention with a commonly used pharmaceutical excipient using, for example, an aerosol spray for inhalation, etc. Alternatively, the benzimidazole derivatives of the present invention in a lyophilized powder form can be administered to the lung using an inhaling device that permits direct contact to the lung.

[0058] To such various formulations, pharmaceutically acceptable carriers such as an isotonic agent, a preservative, a disinfectant, a wetting agent, a buffering agent, an emulsifier, a dispersant, a stabilizer, etc. can be added as needed.

[0059] To these formulations, blending of an antimicrobial agent, a treatment such as filtration through a bacteriaretaining filter, heating, radiation, etc. can be carried out for sterilization. Alternatively, sterile solid formulations can be prepared, which may be used by dissolving or suspending them in an appropriate sterile solution immediately prior to use.

[0060] The dosages of the benzimidazole derivatives of the present invention vary depending on the type of diseases, route of administration, the condition, age, sex, body weight etc. of the patient, but they are generally in the range of about 1 to 500 mg/day/patient for oral administration, and preferably 1 to 300 mg/day/patient. In the case of parenteral administration such as intravenous, subcutaneous, intramuscular, percutaneous, rectal, nasal, eye drop, and inhalation administration, they are about 0.1 to 100 mg/day/patient, and preferably 0.3 to 30 mg/day/patient.

[0061] When the benzimidazole derivatives of the present invention are used as a preventive agent, they can be administered according to a known method depending on each condition.

[0062] As the target diseases for the preventive and/or therapeutic agents of the present invention, there can be mentioned, for example, diseases of respiratory organs such as bronchial asthma, inflammatory/allergic diseases such as allergic rhinitis, atopic dermatitis, and urticaria; diseases of circulatory organs such as sclerosing vascular lesions, intravascular stenosis, disturbances of peripheral circulation, renal failure, and cardiac failure; diseases of bone/cartilage metabolism such as rheumatoid arthritis and osteoarthritis.

50 Examples

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[0063] The present invention will now be explained in more detail with reference to Preparation Examples, Working Examples, and Test Examples. It should be noted, however, that these examples do not limit the scope of the invention in any way.

Reference Example 1. Preparation of 5,6-dimethylbenzimidazole-2-thiol

[0064] To 5,6-dimethylorthophenylene diamine (4.5 g, 33 mmol) in pyridine (40 ml) was added carbon disulfide (40

ml, 0.66 mol). The resulting solution was heated to reflux under stirring for 18 hours, to which was added water, followed by extraction with ethyl acetate. After drying the ethyl acetate phase with anhydrous magnesium sulfate, it was concentrated, and dried under reduced pressure at 80°C for 6 hours to obtain the title compound (4.1 g, yield 70%).

5 Reference Example 2, Preparation of 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl)benzoic acid methyl ester

[0065] To the resulting 5,6-dimethylbenzimidazole-2-thiol (89 mg, 0.50 mmol) in dimethylformamide (2 ml), triethylamine (84 µl, 0.6 mmol) and 2-bromomethyl benzoic acid methyl ester (137 mg, 0.6 mmol) were added. After the resulting solution was stirred at 80°C for 1.5 hours, water was added, followed by extraction with ethyl acetate. After drying the ethyl acetate phase with anhydrous magnesium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 3:1) to obtain the title compound (146 mg, yield 90%). The compound was confirmed by identification of molecular weight using LC-MS. Calculated M = 326.11, measured (M+H)⁺ = 327.2

15 Reference Example 3.

[0066] In a similar manner to Reference Example 2, the following compounds were synthesized. The compounds were confirmed by identification of molecular weight using LC-MS.

20 3-((5,6-dimethylbenzimidazole-2-ylthio)methyl)pyridine-2-carboxylic acid ethyl ester

[0067]

Calculated M = 341.12, found $(M+H)^+ = 342.2$

2-((5,6-dimethylbenzimidazole-2- ylthio)methyl)furane-3-carboxylic acid methyl ester

[0068]

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30 Calculated M = 316.09, found $(M+H)^+ = 317.2$

3-((5,6-dimethylbenzimidazole-2-ytthio)methyl)thiphene-2-carboxylic acid methyl ester

[0069]

Calculated M = 332.07, found $(M+H)^+ = 333.2$

2-(benzimidazole-2-ylthiomethyl)benzoic acid methyl ester

40 [0070]

Calculated M = 298.08, found $(M+H)^+ = 299.2$

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3-(benzimidazole-2-ylthiomethyl)pyridine-2-carboxylic acid ethyl ester

[0071]

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Calculated M = 313.09, found $(M+H)^+ = 314.2$

50 3-(benzimidazole-2-ylthiomethyl)thiophene-2-carboxylic acid methyl ester

[0072]

Calculated M = 304.03, found $(M+H)^+ = 305.2$

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2-(benzimidazole-2-ylthiomethyl)furane-3-carboxylic acid methyl ester

[0073]

5 Calculated M = 288.06, found (M+H)⁺ = 289.2

4-benzimidazole-2-ylthiobutanoic acid methyl ester

[0074]

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Calculated M = 264.09, found $(M+H)^+ = 265.2$

2-((5,6-dichlorobenzimidazole-2-ylthio)methyl)-5-chlorobenzoic acid methyl ester

15 [0075]

Calculated M = 399.96, found $(M+H)^+ = 401.2$

2-(benzimidazole-2-ylthiomethyl)-5-chlorobenzoic acid methyl ester

[0076]

Calculated M = 332.04, found $(M+H)^+ = 333.2$

25 4-((5,6-dimethylbenzimidazole-2-ylthio)butanoic acid ethyl ester

[0077]

Calculated M = 292.12, found $(M+H)^+ = 293.40$

2-((5,6-dichlorobenzimidazole-2-ylthio)methyl)benzoic acid methyl ester

[0078]

35 Calculated M = 366.00, found (M+H)+ = 367.0

2-((5,6-dichlorobenzimidazole-2-ytthio)methyl)pyridine-3-carboxylic acid methyl ester

[0079]

Calculated M = 366.99, found $(M+H)^+ = 368.0$

Example 1 Preparation of compound No. 143

Sodium hydride (11 mg, 0.306 mmol) and 2 ml of tetrahydrofuran was added to a previously dried reaction vessel. To the mixture were added 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl)benzoic acid methyl ester (50 mg, 0.153 mmol) and 1-chloromethylnaphthalene (69 μl, 0.459 mmol), which was then stirred at 60°C for 45 minutes. Water was added thereto, followed by extraction with ethyl acetate. After drying the ethyl acetate phase with anhydrous sodium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 4:1) to obtain 2-((5,6-dimethyl-1-(1-naphthylmethyl)benzimidazole-2-ylthio)methyl)benzoic acid methyl ester (yield 32%).

[0081] To 2-((5,6-dimethyl-1-(1-naphthylmethyl)benzimidazole-2-ylthio)methyl)benzoic acid methyl ester (23 mg, 0.08 mmol) in tetrahydrofuran (1 ml) and methanol (0.5 ml), 4N aqueous sodium hydroxide solution (0.25 ml) was added. After stirring at room temperature for 5 hours, 6N hydrochloric acid was added to stop the reaction, followed by extraction with ethyl acetate. The ethyl acetate phase was washed with saturated saline, and then dried in anhydrous sodium sulfate. The solvent was evaporated under reduced pressure to obtain the title compound (24 mg, yield quantitative).

[0082] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 452.16, found $(M+H)^+ = 453.2$

Example 2.

5 [0083] In a similar manner to Working Example 1, the compounds in Tables 41 to 45 were synthesized using the compounds in Reference Examples 2 or 3 and various halide derivatives. The compounds were confirmed by identification of molecular weight using LC-MS.

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Table 41

Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)
390	406.14	407.2	29
391	422.11	423.2	16
315	417.15	418.2	32
376	406.14	407.2	25
333	417.15	418.2	6
82	416.16	417.2	12
83	416.16	417.2	9
84	416.16	417.2	33
97	432.15	433.2	18
98	432.15	433.2	26
99	432.15	433.2	8
94	470.13	471.2	14
95	470.13	471.2	10
96	470.13	471.2	13
100	486.12	487.2	26
101	486.12	487.2	8
85	420.13	421.2	9
86	420.13	421.0	12
87	420.13	421.2	44
88	436.10	437.2	42
89	436.10	437.2	40 ~
90	436.10	437.2	28
91	480.07	481.0	12
103	427.14	428.2	12
104	427.14	428.2	6
105	427.14	428.2	11
784	434.11	435.2	36

Table 42

Calculated M

468.07

418.14

480.12

494.17

446.13

478.17

452.16

452.16

428.16

458.20

424.12

453.15

430.17

430.17

430.17

430.17

462.16

462.16

492.17

446.13

446.17

446.17 470.06

470.06

470.06

470.06

430.17

Found (M+H)+

469.2

419.2

481.0

495.2

447.2

479.2

453.2

453.0

429.4

459.2

425.2

454.5

431.5

431.5

431.5

431.5

463.0

463.0

493.0

447.0

447.0

447.0

471.0

471.0

471.0

471.0

431.3

Recovery % (overall)

31 40

72

34

19

6

35

30

22

5

25

21

68

52

41

59 25

27.

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75

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37

5			
10			
15			
20			
<i>2</i> 5			

Compound No.

787

112

141

138

135

137

143

142

139

140

63

311

115 116

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118

125

126 128

134

108

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120

121

122

123

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*3*5

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Table 43

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Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)
124	462.16	463.3	67
127	462.16	463.3	62
129	446.17	447.3	47
130	446.17	447.3	40
319	425.12	426.3	30

Table 43 (continued)

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Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)
506	466.17	467.2	16
505	466.17	467.0	14
93	480.07	481.0	45
136	478.17	479.2	60
37	402.14	403.4	25
39	442.03	443.0	51
317	403.14	404.0	56
318	443.03	444.0	46
380	442.14	443.2	51
377	420.15	421.2	34
378	460.04	461.0	30
386	414.10	415.2	37
383	392.12	393.2	30
384	432.01	433.0	29
395	458.11	459.2	23
392	436.13	437.2	15
393	476.02	477.0	15
401	430.08	431.2	50
398	408.10	409.2	20
399	447.99	449.0	7

Table 44

Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)	
544	476.18	377.2	62	
50	418.14	419.2	42	
459	382.08	383.2	65	
402	436.04	437.2	50	
1	388.12	389.0	38	
161	456.05	457.0	54	
81	402.14	403.3	57	
154	444.13	445.0	32	
160	408.10	409.0	72	
159	421.15	422.2	84	
148	482.17	483.5	64	
149	453.15	454.5	71	1
155	459.11	460.0	64	

Table 44 (continued)

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Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)
150	453.15	454.2	36
151	487.11	488.1	62
153	460.10	461.0	69
152	454.15	455.0	62
64	430.08	431.2	85
455	410.11	411.2	17
596	430.14	431.2	56
539	418.17	419.2	20
349	436.10	437.1	50
352	458.09	459.2	74
168	470.06	471.1	57
355	504.02	505.0	26
174	492.05	493.0	89
358	526.01	527.1	38

Table 45

		Table 45		
Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)	
324	493.04	494.2	32	
320	431.08	432.1	15	
147	466.17	467.2	72	
616	490.16	491.2	22	
805	382.17	383.2	52	
804	368.16	369.2	56	
66	438.14	440.2	54	
592	430.14	432.3	5	
811	380.16	382.2	72	
582	436.06	437.1	59	
580	436.06	437.1	59	
584	480.03	483.1	37	
583	480.03	483.0	52	1 .
578	420.09	421.2	30	1
574	416.12	417.2	39	
595	452.12	453.2	22	İ
594	478.14	479.1	23	1
588	432.11	433.1	65	1
587	432.11	433.2	48	
586	432.11	433.1	50	1

Table 45 (continued)

Compound No.	Calculated M	Found (M+H)+	Recovery % (overall)
590	427.10	428.2	24
589	427.10	428.3	17

Example 3. Preparation of compound No. 547

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[0084] Triethylamine (276 µl, 1.98 mmol) and 2-(bromoethyl)benzoic acid t-butyl ester (538 mg, 1.99 mmol) were added to 5,6-dimethylbenzimidazole-2-thiol (236 mg, 1.32 mmol) in 2 ml of dimethylformamide, which was then stirred at 80°C for 3 hours. After the reaction was complete, water was added, followed by extraction with ethyl acetate. After drying the ethyl acetate phase with anhydrous sodium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 3:1) to obtain 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl)benzoic acid t-butyl ester (288 mg, yield 59%).

[0085] 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl)benzoic acid t-butyl ester (30 mg, 0.082 mmol) was dissolved in 3 ml of chloroform, to which triethylamine (17 µl, 0.123 mmol) and benzoyl chloride (14 µl, 0.123 mmol) were sequentially added and the mixture was stirred at room temperature for 2 hours. After the reaction was complete, water was added, followed by extraction with ethyl acetate. After drying the ethyl acetate phase with anhydrous sodium sulfate, it was concentrated, and 2-((5,6-dimethyl-1-(phenylcarbonyl)benzimidazole-2-ylthio)methyl)benzoic acid t-butyl ester was obtained (38 mg, yield quantitative).

[0086] 2-((5,6-dimethyl-1-(phenylcarbonyl)benzimidazole-2-ylthio)methyl)benzoic acid t-butyl ester was dissolved in 1 ml of dichloromethane, to which trifluoroacetic acid (1 ml) was added and the mixture was stirred at room temperature for 6 hours. After the reaction was complete, the solvent was evaporated under reduced pressure and dried overnight to obtain the title compound (33 mg, yield quantitative).

[0087] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 416.12, found $(M+H)^+ = 417.0$

Example 4. Preparation of compound No. 561

[0088] The title compound was obtained in a similar manner to Working Example 3.

[0089] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 452.09, found $(M+H)^+$ = 453.2

Reference Example 4. Preparation of 3-(naphthylmethyl)imidazolo(5,4-b)pyridine-2-thiol

[0090] To 2-amino-3-nitropyridine (1680 mg, 12 mmol) in a dimethylformamide (20 ml), sodium hydride (75 mg, 0.55 mmol) and 1-chloromethylnaphthalene (74 µl, 0.55 mmol) were added. After the resulting solution was stirred at 80°C for 17 hours, water was added thereto, followed by extraction with ethyl ether. After caying the control of the phase with anhydrous magnesium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 4:1) to obtain of naphthylmethyl(3-nitro(2-pyridil))amine (903 mg, yield 27%).

[0091] To naphthylmethyl(3-nitro(2-pyridil))amine (900 mg, 3.2 mmol) in ethanol (40 ml), 90.0 mg of 10% Pd-C was added. After the resulting solution was stirred in a hydrogen atmosphere at 50°C for 8 hours, it was filtered through celite to remove Pd-C. The resulting solution was concentrated to obtain (3-amino(2-pyridil))naphthylmethylamine (860 mg, 3.2 mmol) in ethanol (20 ml), carbon disulfide (6.1 ml, 102 mmol) was added. After the resulting solution was heated to reflux under stirring for 12 hours, it was allowed to stand at room temperature for 5 hours. The precipitate that deposited was filtered, and was washed three times with ethanol (5 ml). It was dried at 80°C under reduced pressure for 5 hours to obtain the title compound (555 mg, yield 56%)

[0092] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 291.08, found $(M+H)^+ = 292.3$

Reference Example 5. Preparation of 3-((2,5-dimethyl)methyl)imidazolo(5,4-b)pyridine-2-thiol

[0093] The title compound was synthesized in a similar manner to Reference Example 4.

[0094] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 269.01, found $(M+H)^+ = 270.2$

5 Example 5. Preparation of compound No. 256

[0095] Using 3-(naphthylmethyl)imidazolo(5,4-b)pyridine-2-thiol (30 mg, 0.1 mmol) obtained in Reference Example 4 in a similar manner to Reference Example 2, 2-((3-(naphthylmethyl)imidazolo(5,4-b)pyridine-2-yithio)methyl)benzoic acid methyl ester was obtained (30 mg, yield 70%).

[0096] The 2-((3-(naphthylmethyl)imidazolo(5,4-b)pyridine-2-thio)methyl)benzoic acid methyl ester (30 mg, 0.068 mmol) thus obtained was subjected to hydrolysis in a similar manner to Example 1 to obtain the title compound (18.3 mg, yield 66%).

[0097] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 425.12, found $(M+H)^+ = 426.1$

Example 6.

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[0098] The compounds in Table 46 were synthesized using the compounds obtained in Reference Examples 4 and 5 and various halide ester derivatives in a similar manner to Example 5.

[0099] The compounds were confirmed by identification of molecular weight using LC-MS.

Table 46

Compound No.	Calculated M	Found (M+H)+	Yield (Overall) %
253	403.14	407.2	67
327	404.13	423.2	46
329	426.12	418.2	58
361	437.10	438.0	52
364	459.08	460.0	66

Table 47

Compound No.	Calculated M	Found (M+H)+	Yield (Overall) %
321	428.13	429.2	27
354	461.10	462.2	20
460	379.14	380.2	19

Table 48

Compound No.	Calculated M	Found (M+H)+	Yield (Overall) %
52	493.15	494.2	12
53	493.15	494.2	11

Example 7. Preparation of compound No. 264

[0100] 4-methyl-2-nitroaniline (913 mg, 6 mmol) was dissolved in acetonitrile (18 ml), to which anhydrous trifluoro-

acetic acid (1.00 ml, 7.2 mmol) was added and the mixture was subjected to reflux for 1.5 hours. After cooling to room temperature, it was concentrated under reduced pressure and dried to obtain 4-methyl-2-nitro trifluoroacetanilide (1.396 g, yield 94%).

[0101] 4-methyl-2-nitro trifluoroacetanilide (1.396 g, 5.63 mmol) was dissolved in dimethylformamide (14 ml), and then potassium carbonate (940 mg, 6.80 mmol) and 1-chloromethylnaphthalene (1.15 g, 6.51 mmol) were sequentially added at room temperature and heated to 100°C. After 1 hour and 40 minutes, 5N aqueous sodium hydroxide solution (7.5 ml) was added and refluxed as it was for 15 minutes. After 15 minutes, it was cooled to room temperature, and water (180 ml) was added and stored at 4°C overnight. The crystals that deposited were filtered and were dried to obtain ((1-naphthyl)methyl)(4-methyl-2-nitro-phenyl)amine (1.587 g, yield 96%).

[0102] To (1-naphthyl)methyl)(4-methyl-2-nitro-phenyl)amine (1.0021 g, 3.43 mmol), ethanol (5 ml) and 1,4-dioxane (5 ml) were added, and 2.058 M aqueous sodium hydroxide solution (1 ml) was further added, and refluxed in an oil bath. After 15 minutes, it was removed from the oil bath, and zinc powder (897 mg, 13.72 mmol) was fed thereto in portions. Then it was refluxed again in the oil bath for 2 hours. After 2 hours, it was concentrated under reduced pressure, and dissolved in ethyl acetate (50 ml), and washed twice with saturated saline (25 ml). After drying with magnesium sulfate, it was concentrated under reduced pressure and dried to obtain a brown oil of ((1-naphthyl)methyl)(2-amino-4-methyl-phenyl)amine (943.1 mg).

[0103] Subsequently, ((1-naphthyl)methyl)(2-amino-4-methyl-phenyl)amine (943.1 mg, 3.59 mmol) was dissolved in ethanol (6.4 ml), to which carbon bisulfide (7 ml, 116 mmol) was added, and then refluxed. After 10 hours, it was returned to room temperature, concentrated under reduced pressure. Ethanol (2 ml) was added to the residue, which was stirred at room temperature for 30 minutes, and was further stirred on ice for 30 minutes. The resulting crystals were filtered, and dried to obtain 1-((1-naphthyl)methyl)-6-methyl-benzimidazole-2-thiol (459.1 mg, yield 44%, 2 steps). [0104] 1-((1-naphthyl)methyl)-6-methyl-benzimidazole-2-thiol (431.1 mg, 1.42 mmol) was dissolved in dimethylformamide (12 ml), to which triethylamine (0.296 ml, 2.12 mmol) and 2-bromomethyl benzoic acid methyl ester (390.1 mg, 1.70 mmol) were added and heated to 80°C. After 5 hours and 50 minutes, triethylamine (0.296 ml, 2.12 mmol) and 2-bromomethyl benzoic acid methyl ester (325 mg, 1.42 mmol) were added, and heated for 1 hour and 10 minutes. Thereafter, it was concentrated under reduced pressure, and dissolved in ethyl acetate (80 ml), washed twice with water (30 ml), and dried in magnesium sulfate. The solvent was concentrated under reduced pressure. The residue was crystalized in ethyl acetate-hexane to obtain 410 mg, and the mother liquor was purified by silica gel column chromatography (hexane: ethyl acetate = 6:1) to recover 87 mg of the same fraction as the crystals, with a total of 497 mg of 2-((1-((1-naphthyl)methyl)-6-methyl-benzimidazole-2-yithio)methyl)benzoic acid methyl ester (yield 78%).

[0105] 2-((1-((1-naphthyl)methyl)-6-methyl-benzimidazole-2-yithio)methyl)benzoic acid methyl ester (497 mg, 1.098 mmol) was dissolved in methanol (10 ml) and tetrahydrofuran (10 ml), to which 4N aqueous lithium hydroxide solution (6.86 ml) was added. After stirring at room temperature for 2 hours and 30 minutes, saturated aqueous citric acid solution (10 ml) was added thereto to stop the reaction, and the mixture was concentrated under reduced pressure to reduce the amount of the solvent to about 1/3, which was dissolved in ethyl acetate (80 ml) and washed five times with water (20 ml). After concentrating the organic layer under reduced pressure, acetonitrile (10 ml) was added to the residue, which was again concentrated under reduced pressure, and the resulting crystals were filtered off and dried to obtain the title compound (439.1 mg, yield 91%).

[0106] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 438.14, found $(M+H)^+ = 439.3$

Example 8. Preparation of compound No. 272

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[0107] In a similar method to working Example 7, the title compound was obtained.
 [0108] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 454.14, found $(M+H)^+ = 455.3$

Example 9. Preparation of compound No. 65

[0109] 2-nitroaniline (829 mg, 6 mmol) and 1-methylindole carboxaldehyde (1242 mg, 7.8 mmol) were dissolved in 20 ml of tetrahydrofuran, to which acetic acid (200 µl) and NaBH(OAc)₃ (5087 mg, 24 mmol) were sequentially added and stirred at room temperature overnight. A saturated aqueous sodium hydrogen carbonate solution was added thereto, followed by extraction with ethyl acetate, dried with anhydrous sodium sulfate, and the solvent was evaporated. After purification by silica gel column chromatography (hexane : ethyl acetate = 95:5), ((1-methylindole-3-yl)methyl)(2-nitrophenyl)amine was obtained (264 mg, yield 18%).

[0110] ((1-methylindole-3-yl)methyl)(2-aminophenyl)amine (264 mg, 0.939 mmol) was dissolved in ethanol (10 ml),

and Pd-C (50 mg, 10% Pd, 0.047 mmol) was added thereto, and stirred in hydrogen atmosphere at room temperature for 6 hours. After the reaction was complete, Pd-C was filtered off and the solvent was evaporated to obtain ((1-methyl-indole-3-yl)methyl)(2-aminophenyl)amine (212 mg, yield 90%).

[0111] ((1-methylindole-3-yl)methyl)(2-aminophenyl)amine (212 mg, 0.845 mmol) was dissolved in pyridine (1 ml), and carbon bisulfide (1 ml, 16.9 mmol) was added thereto. The mixture was refluxed in nitrogen atmosphere for 1 hour. After the solvent was evaporated, it was purified by silica gel column chromatography (hexane: ethyl acetate = 2:1) to obtain ((1-methylindole-3-yl)methyl)benzmidazole-2-thiol (96 mg, yield 39%).

[0112] Sodium hydride (12 mg, 0.342 mmol) and dimethylformamide (2 ml) were added to a previously dried reaction vessel. To the mixture were added ((1-methylindole-3-yl)methyl)benzimidazole-2-thiol (50 mg, 0.171 mmol) and 2-bromomethyl benzoic acid methyl ester (59 mg, 0.257 mmol), and then the mixture was stirred at 60°C for 1 hour. Water was added thereto, followed by extraction with ethyl acetate. After the ethyl acetate phase was dried with anhydrous sodium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 2:1) to obtain 2-((1-((-methylindole-3-yl)methyl)benzimidazole-2-ylthio)methyl)benzoic acid methyl ester (54 mg, yield 74%).

[0113] To 2-((1-((1-methylindole-3-yl)methyl)benzimdazole-2-ylthio)methyl)benzoic acid methyl ester (54 mg, 0.122 mmol) in tetrahydrofuran (2 ml) and methanol (1 ml), 4N aqueous lithium hydroxide solution (0.5 ml) was added. After stirring at room temperature overnight, 6N hydrochloric acid was added to stop the reaction, followed by extraction with ethyl acetate. After washing the ethyl acetate phase with saturated saline, it was dried with anhydrous sodium sulfate. The solvent was evaporated under reduced pressure to obtain the title compound (48 mg, yield 92%).

v [0114] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 427.14, found $(M+H)^+ = 428.2$

Example 10.

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[0115] The compounds in the above Table 47 were synthesized using various halide ester derivatives in a similar manner to Working Example 9. The compounds were confirmed by identification of molecular weight using LC-MS.

Example 11. Preparation of compound No. 51

[0116] Sodium hydride (104 mg, 2.86 mmol) and tetrahydrofuran (16 ml) were added to a previously dried reaction vessel. To the mixture were added 2-(benzimidazole-2-ylthiomethyl)benzoic acid methyl ester (428 mg, 1.43 mmol) and 2-(bromomethyl)benzoic acid t-butyl ester (466 mg, 3.46 mmol), and then the mixture was stirred at 60°C for 50 minutes. Water was added thereto, followed by extraction with ethyl acetate. After the ethyl acetate phase was dried with anhydrous sodium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 3:1) to obtain 2-((1-((2-((t-butyl)oxycarbonyl)phenyl)methyl)benzimidazole-2-ylthio)methyl)benzoic acid methyl ester (495 mg, yield 71%).

[0117] To 2-((1-((2-((t-butyl)oxycarbonyl)phenyl)methyl)benzimidazole-2-yithio)methyl)benzoic acid methyl ester (248 mg, 0.51 mmol), 4N hydrochloric acid in dioxane (1.28 ml, 5.1 mmol) was added, and stirred at room temperature overnight. After the solvent was evaporated, it was dried under reduced pressure to obtain 2-((2-(methoxycarbonyl)phenyl)methylthio)benzimidazolyl)methyl) benzoic acid (220 mg/yield quantitative)

[0118] 2-((2-(methoxycarbonyl)phenyl)methylthio)benzimidazolyl)methyl) benzoic acid (180 mg, 0.42 mmol) was dissolved in chloroform (6 ml), to which HOBT (68 mg, 0.504 mmol), aniline (46 µl, 0.504 mmol), t-butanol (1.2 ml) and EDCI (97 mg, 0.504 mmol) were sequentially added and stirred overnight at room temperature. Water was added thereto, followed by extraction with dichloromethane. After drying with anhydrous sodium sulfate, it was filtered, and the solvent was evaporated. It was purified by silica gel column chromatography (hexane: ethyl acetate = 3:2) to obtain 2-((1-((2-(N-phenylcarbamoyl)phenyl)methylthio)benzimidazole-2-yithio)methyl)benzoic acid methyl ester (86 mg, yield 40%).

[0119] To the thus obtained 2-((1-((2-(N-phenylcarbamoyl)phenyl)methylthio)benzimidazole-2-yithio)methyl)benzoic acid methyl ester (86 mg, 0.169 mmol) in tetrahydrofuran (2 ml) and methanol (1 ml), 4N aqueous lithium hydroxide solution (0.5 ml) was added, and stirred at 60°C for about 2 hours. 6N aqueous hydrochloric acid solution was added to stop the reaction, which was extracted with ethyl acetate. After washing the ethyl acetate phase with saturated saline, it was dried with anhydrous sodium sulfate. The solvent was evaporated under reduced pressure to obtain the title compound (83 mg, yield quantitative).

[6 [0120] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 493.15, found $(M+H)^+ = 494.2$

Example 12.

[0121] In a similar method to Working Example 11, the compounds shown in the above Table 48 were obtained using various benzoic acid ester derivatives. The compounds were confirmed by identification of molecular weight using LC-MS.

Example 13. Preparation of compound No. 619

[0122] Sodium hydride (400 mg, 10.0 mmol) and dimethylformamide (30 ml) were added to a previously dried reaction vessel. To the mixture were added 2-(benzimidazole-2-ylthiomethyl)benzoic acid methyl ester (1500 mg, 5.0 mmol) and bromoacetate t-butyl ester (1463 mg, 7.5 mmol), and the mixture was stirred at 80°C for 2 hours. Water was added thereto, followed by extraction with ether. After the ether phase was dried with anhydrous sodium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 5:1) to obtain 2-(2-((2-(methoxycarbonyl)phenyl)methylthio)benzimidazolyl)acetic acid t-butyl ester (1298 mg, yield 63%).

[0123] To 2-(2-((2-(methoxycarbonyl)phenyl)methylthio)benzimidazolyl)acetic acid t-butyl ester (1290 mg, 3.13 mmol), trifluoroacetic acid (15 ml) was added, and stirred at room temperature overnight. After the solvent was evaporated, it was dried under reduced pressure to obtain 2-(2-((2-(methoxycarbonyl)phenyl)methylthio)benzimidazolyl)acetic acid (715 mg, yield 64%).

[0124] 2-(2-((2-(methoxycarbonyl)phenyl)methylthio)benzimidazolyl)acetic acid (35 mg, 0.1 mmol) was dissolved in tetrahydrofuran (3 ml), to which aniline (11.2 mg, 0.12 mmol) and EDCI (23 mg, 0.12 mmol) were added, and then the mixture was stirred overnight at room temperature. Water was added thereto, followed by extraction with ethyl acetate. After drying with anhydrous sodium sulfate, it was filtered, the solvent was evaporated. The residue was purified by silica gel column chromatography (hexane: ethyl acetate = 3:2) to obtain 2-((1-((N-phenylcarbamoyl)methyl)benzimidazole-2-ylthio)methyl)benzoic acid methyl ester (27.5 mg, yield 64%).

[0125] 2-((1-((N-phenylcarbamoyl)methyl)benzimidazole-2-yithio)methyl)benzoic acid methyl ester (20 mg, 0.046 mmol) thus obtained was subjected to hydrolysis as in Working Example 1 to obtain the title compound (6.9 mg, yield 36%).

[0126] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 417.11, found $(M+H)^+ = 418.0$

Example 14

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[0127] In a similar method to Example 13, the compounds shown in the above Table 49 were obtained using various aniline derivatives.

[0128] The compounds were confirmed by identification of molecular weight using LC-MS.

Table 49

Compound No.	Calculated	M Found (M+H)+	Yield (Overall) %
622	431.13	432.3	5
621	431.13	432.3	5
620	431.13	432.3	21
637	447.13	448.2	13
636	117.13	448.1	23
635	447.13	448.3	44
642	442.11	443.2	27
657	467.13	488.1	19

Table 50

Compound No.	Calculated M	Found (M+H)+	Yield (Overall) %
765	457.15	458.2	5
767	457.15	458.2	32

Table 51

Compound No.	Calculated M	Found (M+H)+	Yield (Overall) %
866	434.13	435.2	76
869	456.11	457.3	83
904	468.09	469.1	52
937	436.15	437.2	61

Table 52

Compound No.	Calculated M	Found (M+H)+	Yield (Overall) %
953	476.18	477.2	36
985	428.18	429.2	67
977	400.15	401.4	2

Reference Example 6, Preparation of 2-((1-(2-hydroxyethyl)-5,6-dimethylbenzimidazole-2-ylthio)]methyl)benzoic acid methyl ester

[0129] To 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl) benzoic acid methyl ester (326 mg, 1 mmol) obtained in Reference Example 2 in dimethylformamide, potassium carbonate (207 mg, 1.5 mmol) and 2-bromoethanol (150 mg, 1.2 mmol) were added, and the resulting solution was stirred at 80°C for 12 hours. After the reaction was complete, it was extracted with ether and the solvent was evaporated. The residue was purified by a flash column chromatography (hexane: ethyl acetate = 4:1) to obtain the thegitle compound (248 mg; yield:67%).

[0130] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 370.14, found $(M+H)^+ = 371.2$

Example 15, Preparation of compound No. 736

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[0131] To 2-((1-(2-hydroxyethyl)-5,6-dimethylbenzimidazole-2-yithio)methyl)benzoic acid methyl ester (45 mg, 0.23 mmol) in N-methylmorpholine (3 ml), Pph₃ (62 mg, 0.24 mmol) and DEAD (10.6 ml, 40% in toluene, 0.24 mmol) were added and the mixture was stirred at room temperature. After 10 minutes, phenol (11.3 mg, 0.12 mmol) was added thereto, which was stirred at room temperature for 12 hours. The solvent was evaporated and the residue was purified by thin layer chromatography (hexane: ethyl acetate = 1:1) to obtain 2-((5,6-dimethyl-1-(2-phenoxyethyl)benzimidazole-2-yithio)methyl)benzoic acid methyl ester (44 mg, yield 81%).

[0132] Using 2-((5,6-dimethyl-1-(2-phenoxyethyl)benzimidazole-2-ylthio)methyl)benzoic acid methyl ester (35 mg, 0.078 mmol) in a similar method to Example 1, a hydrolysis reaction was carried out to obtain the title compound (31 mg, yield 94%). The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 432.15, found $(M+H)^+ = 433.2$

Example 16.

[0133] In a similar method to Example 15, the compounds shown in the above Table 50 were obtained using various phenol derivatives.

[0134] The compounds were confirmed by identification of molecular weight using LC-MS.

Example 17.

Preparation of compound No. 825

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[0135] To an ester (33 mg, 0.075 mmol) of compound No. 68 obtained in Example 2 in dichloromethane, 50 to 60% m-chloroperbenzoic acid (26 mg, 0.083 mmol) was added while cooling on ice. After the resulting solution was stirred on ice for 2 hours, a saturated sodium hydrogen carbonate solution was poured and the solution obtained was extracted with chloroform. After washing the chloroform phase with water, it was concentrated and the residue was purified by thin layer chromatography (hexane: ethyl acetate = 1:1) to obtain 2-(((5,6-dimethyl-1-(1-naphthylmethyl)benzimidazole-2-yl)sulfinyl)methyl)benzoic acid methyl ester (7.1 mg, yield 21%).

[0136] In a manner similar to Example 1, this was subjected to hydrolysis to obtain the title compound (5.2 mg, yield 76%).

[0137] The compound was confirmed by identification of molecular weight using LC-MS.

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Calculated M = 440.12, found $(M+H)^+ = 441.3$

Example 18. Preparation of compound No. 869

25 [0138] To an ester (39 mg, 0.094 mmol) of compound No. 37 obtained in Example 2 in dichloromethane (5 ml), 50 to 60% m-chloroperbenzoic acid (64 mg, 0.374 mmol) was added while cooling on ice. After the resulting solution was stirred at room temperature for 4 hours, a saturated sodium hydrogen carbonate solution was poured and the solution obtained was extracted with chloroform. After washing the chloroform phase with water, it was concentrated and the residue was purified by flash layer chromatography (hexane: ethyl acetate = 5:1) to obtain 2-(((1-((2,5-dimethylphe-30 myl)methyl)benzimidazole-2-yl)sulfonyl)methyl)benzoic acid methyl ester (37 mg, yield 87%).

[0139] In a manner similar to Example 1, 2-(((1-((2,5-dimethylphenyl)methyl)benzimidazole-2-yl)sulfonyl)methyl)benzoic acid methyl ester (64 mg, 0.14 mmol) was subjected to hydrolysis to obtain the title compound (53 mg, yield 87%).

[0140] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 434.13, measured $(M+H)^+ = 435.2$

Example 19.

[0141] In a manner similar to Example 18, the compounds shown in the above Table 51 were synthesized using the esters of the compounds obtained in Working Example 2. The compounds were confirmed by identification of molecular weight using LC-MS.

Example 20. Preparation of compound No. 952

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[0142] To 5,6-dimethylbenzimidazole-2-thiol (713 mg, 4 mmol) in dimethylformamide (10 ml), triethylamine (836 µl, 6 mmol) and 2-bromomethylbenzonitrile (1176 mg, 6 mmol) were added. After stirring at 80°C overnight, water was added to the mixture, followed by extraction with ethyl acetate. After the ethyl acetate phase was dried with anhydrous sodium sulfate, it was concentrated and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 3:2) to obtain 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl)benzenecarbonitrile (1159 mg, yield 99%).

[0143] Sodium hydride (178 mg, 4.90 mmol) and tetrahydrofuran (30 ml) were added to a previously dried reaction vessel. To the mixture were added 2-((5,6-dimethylbenzimidazole-2-ylthio)methyl)benzenecarbonitrile (719 mg, 2.45 mmol) and 2,5-dichlorobenzyl chloride (543 µl, 4.90 mmol), and the mixture was stirred at 60°C for 40 minutes. Water was added thereto, followed by extraction with ethyl acetate. After the ethyl acetate phase was dried with anhydrous sodium sulfate, it was concentrated, and the residue was purified by silica gel column chromatography (hexane: ethyl acetate = 3:1) to obtain 2-((1-((2,5-dimethylphenyl)methyl)-5,6-dimethylbenzimidazole-2-ylthio)methyl)benzenecarbonitrile (370 mg, yield 37%).

[0144] 2-((1-((2,5-dimethylphenyl)methyl)-5,6-dimethylbenzimidazole-2-ylthio)methyl)benzenecarbonitrile (165

mg, 0.401 mmol) was dissolved in toluene (3 mf), to which Me_3SnN_3 (124 mg, 0.602 mmol) was added, and refluxed in nitrogen atmosphere overnight. After the reaction was complete, the solvent was evaporated, and the residue was purified by silica gel column chromatography (dichloromethane : methanol = 19:1) to obtain the title compound (75 mg, yield 41%).

5 [0145] The compound was confirmed by identification of molecular weight using LC-MS.

Calculated M = 454.19, found $(M+H)^+ = 455.2$

Example 21.

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[0146] In a manner similar to Example 20, the compounds shown in the above Table 52 were obtained.

[0147] The compounds were confirmed by identification of molecular weight using LC-MS.

Example 22. Preparation of recombinant human mast cell chymase

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[0148] Recombinant pro-type human mast cell chymase was prepared according to the method reported by Urada et al. (Journal of Biological Chemistry 266: 17173, 1991). Thus, a culture supernatant of the insect cell (Tn5) infected with a recombinant baculovirus containing cDNA encoding human mast cell chymase was purified by heparin Sepharose (Pharmacia). After it was further activated by the method reported by Murakami et al. (Journal of Biological Chemistry 270: 2218, 1995), it was purified with heparin Sepharose to obtain an activated human mast cell chymase.

Example 23. Determination of the activity of inhibiting recombinant human mast cell chymase

[0149] After a DMSO solution (2 μl) containing the compound of the present invention was added to 50 μl of buffer A (0.5-3.0 M NaCl, 50 mM Tris-HCl, pH 8.0) containing 1-5 ng of the activated human mast cell chymase obtained in working Example 22, 50 μl of buffer A containing, as a substrate, 0.5 mM succinyl-alanyl-histidyl-prolylphenylalanyl-paranitroanilide (Bacchem) was added thereto and the mixture was allowed to react at room temperature for 5 minutes. Changes in absorbance at 405 nm with time were measured to evaluate the inhibitory activity.

[0150] As a result, IC50 = not smaller than 1 nM and less than 10 nM was observed in compounds No. 63, 64, 65, 143, 174, 256, 264, 272, 311, 354, 319, 349, 358, 395, 401, and 402, and IC50 = not smaller than 10 nM and not greater than 100 nM was observed in compounds No. 37, 50, 84, 115, 117, 119,, 121, 123, 130, 147, 168, 256, 320, 321, 324, 352, 355, 364, 380, 392, 398, 444, 455, 459, 460, 506, 863, 866, and 869.

[0151] As hereinabove described, the benzimidazole derivatives of the present invention exhibit a potent chymase inhibitory activity. Thus, it was revealed that the benzimidazole derivatives of the present invention are clinically applicable inhibitory substances for human chymase activity and can be used for prevention and/or therapy of various diseases in which human chymase is involved.

Example 24. Manufacture of tablets

40 [0152] Tablets comprising, per tablet, the following were manufactured:

Compound (No. 37)	50 mg
Lactose	230 mg
Potato starch	80 mg
Polyvinylpyrrolidone	11 mg
Magnesium stearate	5 mg

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[0153] The compound of the present invention (the compound in Working Example 2), lactose and potato starch were mixed, and the mixture was evenly soaked in 20% polyvinylpyrrolidone in ethanol. The mixture was filtered through a 20 nm mesh, dried at 45°C, and filtered again through a 15 nm mesh. Granules thus obtained were mixed with magnesium stearate and were compressed into tablets.

Industrial Applicability

[0154] The thiobenzimidazole derivatives of the present invention and the medically acceptable salts thereof exhibit a potent activity of inhibiting human chymase. Thus, said thiobenzimidazole derivatives and the medically acceptable salts thereof can be used, as a human chymase inhibitor, as clinically applicable preventive and/or therapeutic agents for inflammatory diseases, allergic diseases, diseases of respiratory organs, diseases of circulatory organs, or diseases of bone/cartilage metabolism.

Claims

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1. A thiobenzimidazole derivative represented by the following formula (1):

wherein,

 R^1 and R^2 , simultaneously or independently of each other, represent a hydrogen atom, a halogen atom, a trihalomethyl group, a cyano group, a hydroxy group, an alkyl group having 1 to 4 carbons or an alkoxy group having 1 to 4 carbons, or R^1 and R^2 together form -O-CH₂-O-, -O-CH₂-CH₂-O- or -CH₂-CH₂-CH₂-, in which the carbons may be substituted with one or a plurality of alkyl groups having 1 to 4 carbons;

A represents a single bond, a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, a substituted or unsubstituted arylene group having 6 to 11 carbons, or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, in which the substituent represents a halogen atom, OH, NO₂, CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a linear or branched alkylthio group having 1 to 6 carbons, a linear or branched alkylsulfonyl group having 1 to 6 carbons, a linear or branched acyl group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a trihalomethyl group, a trihalomethoxy group, a phenyl group, an oxo group, or a phenoxy group that may be substituted with one or more halogen atoms, and in which the substituents may be independently substituted at any one or more sites of the ring or the alkylene group;

E represents COOR³, SO₂R³, SO₂NHR³, a tetrazole group, a 5-oxo-1,2,4-oxadiazole group or a 5-oxo-1,2,4-thiadiazole group in which R³ represents a hydrogen atom, or a linear or branched alkyl group having 1 to 6 carbons;

G represents a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons that may be interrupted with one or a plurality of O, S, SO_2 , and NR^3 , in which R^3 is as defined above and the substituent represents a halogen atom, OH, NO_2 , CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a trihalomethyl group, a trihalomethoxy group, a phenyl group, or an oxo group; m represents an integer of 0 to 2;

when m is 0 and A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, then J represents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted aryl group having 7 to 9 carbons, a substituted aryl group having 10 to 11 carbons, a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring;

when m is 0 and A is a substituted or unsubstituted arylene group having 6 to 11 carbons or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, then J represents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted aryl group having 6 to 11 carbons, or a sub-

stituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring; or

when m is 0 and A is a single bond or when m is 1 or 2, then J represents a substituted or unsubstituted, linear, cyclic or branched alkyl group having 1 to 6 carbons, a substituted or unsubstituted aryl group having 6 to 11 carbons, or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, in which the substituent represents a halogen atom, OH, NO₂, CN, a linear or branched alkyl group having 1 to 6 carbons, a linear or branched alkoxy group having 1 to 6 carbons (the substituents may be joined to each other at adjacent sites via an acetal bond), a linear or branched alkylthio group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a linear or branched acylamino group having 1 to 6 carbons, a substituted or unsubstituted anilide group, a trihalomethyl group, a trihalomethoxy group, a phenyl group, an oxo group, a COOR³ group, or a phenoxy group that may be substituted with one or more halogen atoms, and in which the substituents may be independently substituted at any one or more sites of the ring or the alkylene group; and

X represents CH or a nitrogen atom:

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or a medically acceptable salt thereof (hereinafter referred to as "the thiobenzimidazole derivative of the present invention").

- 2. The thiobenzimidazole derivative according to claim 1 characterized in that, in the above formula (1), A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, a substituted or unsubstituted arylene group having 6 to 11 carbons, or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
- 25 3. The thiobenzimidazole derivative according to claim 1 or 2 characterized in that, in the above formula (1), A is a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
- 4. The thiobenzimidazole derivative according to any one of claims 1, 2, and 3 characterized in that, in the above formula (1), m is 1, or a medically acceptable salt thereof.
 - 5. The thiobenzimidazole derivative according to any one of claims 1, 2, and 3 characterized in that, in the above formula (1), m is 2, or a medically acceptable salt thereof.
- 35 6. The thiobenzimidazole derivative according to any one of claims 1, 2, and 3 characterized in that, in the above formula (1), m is 0, A is a substituted or unsubstituted, linear or branched alkylene group having 1 to 6 carbons, and J is a substituted or unsubstituted aryl group having 7 to 9 carbons, a substituted aryl group having 10 to 11 carbons, or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
 - 7. The thiobenzimidazole derivative according to any one of claims 1, 2, and 3 characterized in that, in the above formula (1), m is 0, A is a substituted or unsubstituted arylene group having 6 to 11 carbons or a substituted or unsubstituted heteroarylene group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, and J is a substituted or unsubstituted aryl group having 6 to 11 carbons or a substituted or unsubstituted heteroaryl group having 4 to 10 carbons that may contain one or a plurality of oxygen, nitrogen and sulfur atoms on the ring, or a medically acceptable salt thereof.
 - The thiobenzimidazole derivative according to any one of claims 1 to 7 characterized in that, in the above formula

 (1), G is -CH₂-, -CH₂-CH₂-, -CH₂CO-, -CH₂CH₂O-, -CH₂CONH-, -CO-, -SO₂-, -CH₂SO₂-, -CH₂SO₂-, -CH₂SO₃-, -CH₂CH₂S-, or a medically acceptable salt thereof.
 - 9. The thiobenzimidazole derivative according to any one of claims 1 to 8 characterized in that, in the above formula (1), R¹ and R² simultaneously represent a hydrogen atom, a halogen atom, an alkyl group having 1 to 4 carbons or an alkoxy group having 1 to 4 carbons, or R¹ and R², independently of each other, represent a hydrogen atom, a halogen atom, an alkyl group having 1 to 4 carbons, an alkoxy group having 1 to 4 carbons, a trihalomethyl group, a cyano group, or a hydroxy group, or a medically acceptable salt thereof.
 - 10. The thiobenzimidazole derivative according to any one of claims 1 to 9 characterized in that, in the above formula

- (1), E represents COOH or a tetrazole group, or a medically acceptable salt thereof.
- The thiobenzimidazole derivative according to any one of claims 1 to 10 characterized in that, in the above formula
 X represents CH, or a medically acceptable salt thereof.
- 12. The thiobenzimidazole derivative according to any one of claims 1 to 11 characterized by having an activity of inhibiting human chymase, or a medically acceptable salt thereof.
- 13. A pharmaceutical composition comprising at least one thiobenzimidazole derivative according to any one of claims
 1 to 12 or a medically acceptable salt thereof and a pharmaceutically acceptable carrier.
 - 14. The pharmaceutical composition according to claim 13 which is a preventive and/or therapeutic agent of a disease.
 - 15. A preventive and/or therapeutic agent according to claim 14 wherein said disease is an inflammatory disease, an allergic disease, a disease of respiratory organs, a disease of circulatory organs, or a disease of bone/cartilage metabolism.

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NTERNATIONAL SEARCH REPORT

International application No.
PCT/JP99/03799

2.4384.734.765

A. CLASSIFICATION OF SUBJECT MATTER Int.Cl* C07D235/28, A61K31/415				
According t	to International Patent Classification (IPC) or to both n	ational classification and IPC		
B. FIELD	S SEARCHED			
Minimum d Int	Minimum documentation searched (classification system followed by classification symbols) Int.Cl ⁶ C07D235/28, A61K31/415			
Documenta	Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched			
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) CAPLUS (STN), REGISTRY (STN)				
C. DOCU	MENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document, with indication, where ap		Relovant to claim No.	
X	US, 5124336, A (Laboratoires 23 June, 1992 (23. 06. 92)	UPSA),	1, 2, 8-15	
A	& FR, 2658511, A		3-7	
x	J. Med. Chem., 36(9), 1175-8	7 (1993)	1, 2, 8-15	
x	JP, 5-155858, A (Laboratories UPSA), 22 June, 1993 (22. 06. 93) & US, 5021443, A & PR, 2658511, A & CA, 2035710, A & BP, 442820, A		1, 2, 8-15	
X	Chem. Abs. <u>85</u> , 172661 (1976)		1, 2, 8-12	
Porth	er documents are listed in the continuation of Box C.	See patent family annex.		
Special entegories of cited documents: A document defining the general state of the art which is not considered to be of particular relevance. E carlier document but published on or after the international filing date or priority described to earther document with the application but cited to understand the priority during the invention cannot be considered to involve an inventive step when the document in laters alone occurrent which may throw doubts on priority chain(s) or which is cited to establish the publication dute of another citation or other special reason (as specified) O document referring to an end disclosure, use, exhibition or other means P document published prior to the international filing date but later than the priority date claimed Date of the actual completion of the international search O Ctober, 1999 (07. 10. 99) T bete document published after the international search to document in the name patent family Date of mailing of the international search report 19 October, 1999 (19. 10. 99)			ion but cited to understand contion issued invention cannot be if to involve an inventive step simed invention cannot be then the document in occurrent, such combination at mily	
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